

## PART 2: TYPES OF BONDS

A chemical bond is an attraction between two or more atoms. Every type of chemical bond is based on the electronegativity of the atoms involved in the bonds. Types of bonds/interactions include:

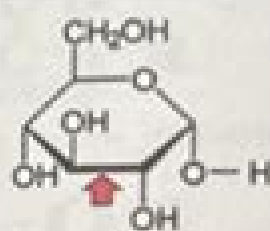
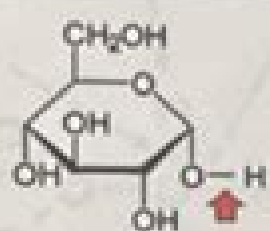
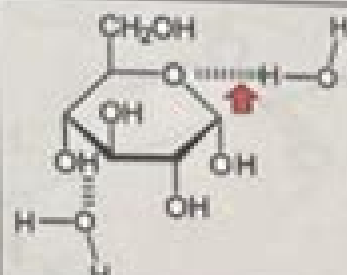

**Ionic Bonds:** form when one atom steals an electron from another, resulting in two oppositely charged ions. The ions are electrostatically attracted to each other (opposites attract).

**Covalent bonds:** form when two atoms share electrons.

**Hydrogen bonds:** are a type of electrostatic attraction between oppositely charged dipoles. Dipoles are not full charges.

1. Fill in the bond names below:

Bond Name:

2. The hydrogen bond example above is drawn with a dashed line. Hydrogen bonds should always be demonstrated with a dashed line. Draw three water molecules below and the hydrogen bonds occurring between them.

# Chemical Bonds Packet Section 2 Types Of Bonds

## Answers

**N Noddings**



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