

# Naming Chemical Compounds Worksheet

Name the following ionic compounds:

- 1)  $\text{NaBr}$  \_\_\_\_\_
- 2)  $\text{CaO}$  \_\_\_\_\_
- 3)  $\text{Li}_2\text{S}$  \_\_\_\_\_
- 4)  $\text{MgBr}_2$  \_\_\_\_\_
- 5)  $\text{Pu}(\text{OH})_3$  \_\_\_\_\_
- 6)  $\text{Hg}(\text{CN})_2$  \_\_\_\_\_
- 7)  $\text{Mo}(\text{C}_2\text{H}_3\text{O}_2)$  \_\_\_\_\_
- 8)  $\text{Cr}(\text{CrO}_4)_3$  \_\_\_\_\_
- 9)  $\text{WO}_2$  \_\_\_\_\_
- 10)  $\text{Mg}(\text{ClO})_2$  \_\_\_\_\_

Write the formulas for the following ionic compounds:

11. potassium dichromate \_\_\_\_\_
12. gold(III) oxide \_\_\_\_\_
13. aluminum periodate \_\_\_\_\_
14. barium nitrite \_\_\_\_\_
15. silver carbonate \_\_\_\_\_
16. lithium sulfite \_\_\_\_\_
17. zinc hydrogen carbonate \_\_\_\_\_
18. iron(III) hydroxide \_\_\_\_\_
19. ammonium phosphate \_\_\_\_\_
20. copper(II) bromite \_\_\_\_\_
  
- 21)  $\text{SO}_3$  \_\_\_\_\_
- 22)  $\text{N}_2\text{S}$  \_\_\_\_\_
- 23)  $\text{BF}_3$  \_\_\_\_\_
- 24)  $\text{P}_2\text{Br}_4$  \_\_\_\_\_
- 25)  $\text{SiO}_2$  \_\_\_\_\_
- 26)  $\text{SF}_6$  \_\_\_\_\_
- 27)  $\text{NO}_2$  \_\_\_\_\_
- 28) nitrogen trichloride \_\_\_\_\_
- 29) dinitrogen trioxide \_\_\_\_\_
- 30) phosphorus pentafluoride \_\_\_\_\_
- 31) diboron tetrahydride \_\_\_\_\_

# Chemical Formulas And Nomenclature Answers

**Lloyd McKinley**



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**Dental State Board Questions and Answers** Rudolph Max Goepp, 1922

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**Inorganic Chemical Nomenclature** Lloyd McKinley, 1947

**School Science and Mathematics**, 1918

*Journal of Chemical Education*, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers

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