#### Determining Chemical Formulas Lab Report Sheet

#### Date: 3/11/2023

Name: kb bohm

#### Lab Partner(s): Jared

#### Purpose:

Prepare for the lab by writing a brief paragraph on the Purpose. Use complete sentences, avoiding the use of "I" or "we". Be sure to address the following items:

- What is the main topic/concept being investigated today?
- 2. How will you investigate this topic/concept?
- 3. Write the main definitions and equations needed for today's experiment.

The main topic of this investigation is determining empirical formula of a hydrated salt sample, a hydrated salt sample will be heated, while this sample is being heated there is going to be an evaporation of water, leaving just salt behind, when there is just salt left behind this is called anhydrous salt, quantitative analysis will be perform during the entire process, recording the initial mass of the Cuso4 \* XH2O and the final mass of Cuso4, also determining how much water was evaporated during the process, after the heating process water will be added.

anhydrous salt- salt without any water molecules

mass % of Z = mass of Z / mass of compound × 100 %

Percent error= [True value-Experimental value] / True value × 100 %

#### Data:

Table 1: Data for unknown sample undergoing dehydration reaction.

Mass of foil pan:	2.367 g	

# **Chemical Formulas Lab Report**

**JR Anderson** 

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