

Chapter 3

Chemical Foundations: Elements, Atoms, and Ions

I. Matching

Match the description in Column B with the correct term in Column A. Write the letter in the blank provided. Each term matches with only one description, so be sure to choose the best description for each term.

Column A

- _____ 1. electron
- _____ 2. nucleus
- _____ 3. isotopes
- _____ 4. atomic number
- _____ 5. mass number
- _____ 6. ion
- _____ 7. neutron
- _____ 8. cation
- _____ 9. proton
- _____ 10. anion

Column B

- A. sum of the number of protons and the number of neutrons in the nucleus of an atom
- B. positively charged subatomic particle
- C. dense, positively charged center of atom
- D. subatomic particle with zero charge
- E. atoms with same number of protons but different numbers of neutrons
- F. negatively charged ion
- G. number of protons in the nucleus of an atom
- H. atom with a charge
- I. positively charged ion
- J. negatively charged subatomic particle

II. Multiple Choice

Choose the one best answer and write its letter in the blank.

- _____ 11. Which of the following is a compound?

- a) Fe b) CO c) O₂ d) K

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