

Name _____ Date _____

9-3 Section Review

1. Name these binary molecular compounds

- a. PCl_3 _____
- b. SF_6 _____
- c. OF_2 _____
- d. Cl_2O_8 _____
- e. SO_3 _____
- f. CCl_4 _____
- g. PBr_3 _____
- h. P_4S_3 _____

2. Write the formulas for the following binary molecular compounds

- i. nitrogen trifluoride _____
- j. disulfur dichloride _____
- k. dinitrogen tetraoxide _____
- l. sulfur dibromide _____
- m. dinitrogen pentaoxide _____
- n. xenon tetrafluoride _____
- o. dioxygen difluoride _____
- p. diphosphorous trioxide _____

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