

## Answers – Naming Chemical Compounds

*Name the following chemical compounds:*

- |     |  |                        |
|-----|--|------------------------|
| 1)  | NaBr   | sodium bromide         |
| 2)  | Ca(C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> ) <sub>2</sub> | calcium acetate        |
| 3)  | P <sub>2</sub> O <sub>5</sub>                                  | diphosphorus pentoxide |
| 4)  | Ti(SO <sub>4</sub> ) <sub>2</sub>                              | titanium(IV) sulfate   |
| 5)  | FePO <sub>4</sub>  | iron(III) phosphate    |
| 6)  | K <sub>3</sub> N   | potassium nitride      |
| 7)  | SO <sub>2</sub>  | sulfur dioxide         |
| 8)  | CuOH   | copper(I) hydroxide    |
| 9)  | Zn(NO <sub>2</sub> ) <sub>2</sub>                              | zinc nitrite           |
| 10) | V <sub>2</sub> S <sub>3</sub>                                  | vanadium(III) sulfide  |

*Write the formulas for the following chemical compounds:*

- |     |                                |   |
|-----|--------------------------------|---|
| 11) | silicon dioxide                | SiO <sub>2</sub>                                |
| 12) | nickel (III) sulfide           | Ni <sub>2</sub> S <sub>3</sub>                  |
| 13) | manganese (II) phosphate       | Mn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> |
| 14) | silver acetate                 | AgC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>  |
| 15) | diboron tetrabromide           | B <sub>2</sub> Br <sub>4</sub>                  |
| 16) | magnesium sulfate heptahydrate | MgSO <sub>4</sub> ·7H <sub>2</sub> O            |
| 17) | potassium carbonate            | K <sub>2</sub> CO <sub>3</sub>                  |
| 18) | ammonium oxide                 | (NH <sub>4</sub> ) <sub>2</sub> O               |
| 19) | tin (IV) selenide              | SnSe <sub>2</sub>                               |
| 20) | carbon tetrachloride           | CCl <sub>4</sub>                                |

# Chemical Names And Formulas Lab Answers

**Kenneth D. Kerri**



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