

COMMON TYPES OF CHEMICAL REACTIONS

SYNTHESIS

Multiple reactants, which can be simple elements or compounds, combine together to form a single compound.



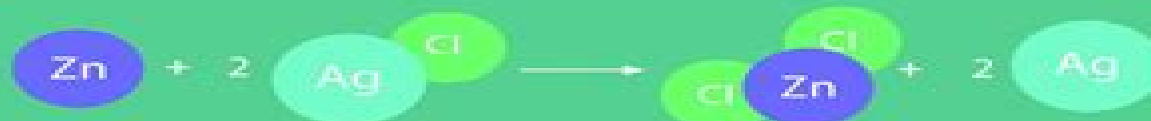
DECOMPOSITION

A compound breaks down into two or more simpler substances. Decomposition reactions are classified into thermal, electrolytic, and photo.



SINGLE-REPLACEMENT

One element is substituted for another element in a compound, generating a new compound and a pure element.



DOUBLE-REPLACEMENT

Two ionic compounds exchange cations or anions to form two new compounds. Forming a precipitate can help drive the reaction to the right.



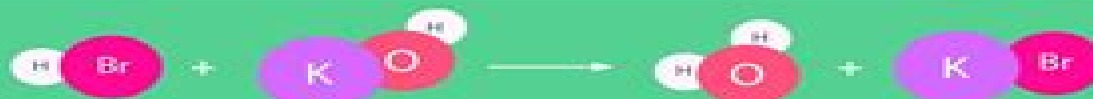
COMBUSTION

A fuel, typically a hydrocarbon, reacts with oxygen gas to form carbon dioxide and water, which generates heat and light.



NEUTRALIZATION

Acid and base combine to neutralize each other in aqueous solutions, usually resulting in a salt and water.



Chemical Reactions Chem Think Answers

Rachel S Tattersall



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Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions **Prentice Hall Chemistry** ,2000

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