

Chemistry Worksheets Class 11 on Chapter 4 Chemical Bonding and Molecular Structure - Set 3

Q-1: Which of the	following	possesses an	expanded	octet?
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- a) IF,
- b) SF.
- c) BrF_n
- d) All of the above

Q-2: Strongly electronegative element B contrasts with strongly electropositive element A. Both are divalent. The resulting compound would be

- a) A-B
- b) A2*B2-
- c) A2-B2"
- d) A-B

Q-3: Which of the following compounds shows the highest lattice energy?

- a) CsCl
- b) KCI
- c) RbCl
- d) NaCl

Q-4: Which one of the following molecules is formed by p-p overlapping?

- a) I_2
- b) HBr
- c) H₂S
- d) PH_a

Q-5: Which of the following H-bonds has the greatest impact on a molecule's physical characteristics?

- a) Intramolecular H-bond
- b) Intermolecular H-bond
- c) Both of the above
- d) All H-bonds

Q-6: State the crucial conditions that must be met for a molecule to undergo hybridisation.

Q-7: Describe why CH₄ has a tetrahedral geometry rather than a square planar geometry with a carbon atom in the centre and four H atoms at each corner.

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