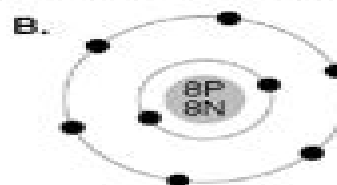
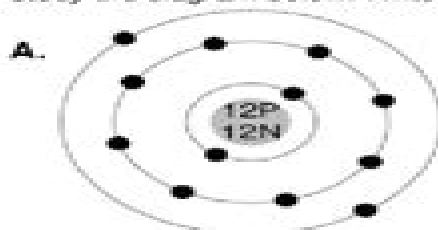


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# IPS Unit 9 – Chemical Bonds

## Types of Bonds Worksheet

**Directions:** Study the diagram below. Write your answers to the questions in the spaces provided.



1. If atom A loses electrons to atom B,
  - a. how many electrons will atom A lose? \_\_\_\_\_
  - b. how many electrons will atom B gain? \_\_\_\_\_
  - c. what will be the charge of atom A? \_\_\_\_\_
  - d. what will be the charge of atom B? \_\_\_\_\_
  - e. what will be the total charge of the compound formed?  
\_\_\_\_\_
  - f. what type of bond will form?  
\_\_\_\_\_
2. Explain why an element's charge is related to the group on the periodic table to which it belongs.  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_

**Directions:** Complete the table comparing ionic compounds and covalent compounds.

Characteristic	Ionic compounds	Covalent compounds
How the compound is formed		
Smallest particle		
Usual state at room temperature		

# Chemical Bonding 8 3 1answer Key

**Didier Musso**



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