

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Chemical Equations

**Vocabulary:** Avogadro's number, chemical equation, chemical formula, chemical reaction, coefficient, combination, combustion, conservation of matter, decomposition, double replacement, molar mass, mole, molecule, product, reactant, single replacement, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and an equal weight is placed on the other pan.

What would happen if you lit up the candle and waited for a while? \_\_\_\_\_

\_\_\_\_\_

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by an equal weight.

In this situation, what would happen if you lit up the candle and waited? \_\_\_\_\_

\_\_\_\_\_

### Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations Gizmo™*, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the **Reactants** box and "H<sub>2</sub>O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.



1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.
- A. How many hydrogen atoms are on the **Reactants** side? \_\_\_\_\_ **Products** side? \_\_\_\_\_
- B. How many oxygen atoms are on the **Reactants** side? \_\_\_\_\_ **Products** side? \_\_\_\_\_
2. Based on what you see, is this equation currently balanced? \_\_\_\_\_

# Chemical Equations Gizmo Answer Key

**Learning Hub Publishing**



## **Chemical Equations Gizmo Answer Key:**

**Chemistry Equations & Answers** Speedy Publishing,2014-08-01 One of the most important parts of learning chemistry is simply knowing all of the equations and formulas that are used in it It is important to know the exact formatting of these equations and most classes will require a student to know them for exams It is a good idea to learn these equations and formulas with the use of a study pamphlet The pamphlet can condense all of the information so a student can memorize the equations and formulas while studying *Chemistry Equations & Answers* Speedy Publishing,2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations

**Chemistry Equations & Answers** Pamphlet Master,2014-10-12 This Chemistry Equations Answers study guide is created by Pamphlet Master for students everywhere This tool has a comprehensive variety of college and graduate school topics subjects which can give you what it takes to achieve success not only in school but beyond Included in the pamphlet are Chemical Formula and Equations What is a Chemical Formula Chemical Formula and Equations Subscripts What Is A Balanced Equation How Do We Balance The Equation What About These Halves Examples of Balancing Chemical Equations

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not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of  $C_2H_2$  and  $O_2$  to give  $CO_2$  and  $H_2O$  Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms Stormy weather The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet

The Ultimate Chemical Equations Handbook, Teacher Edition Jane D. Smith, George R. Hague, 2011      The Ultimate Chemical Equations Handbook, Student Edition Jane D. Smith, George R. Hague, 2011      Tests and quizzes, 1987      **General Chemistry Key Concise Review Guide Sixth Edition** Darrell D. Ebbing, 1999

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