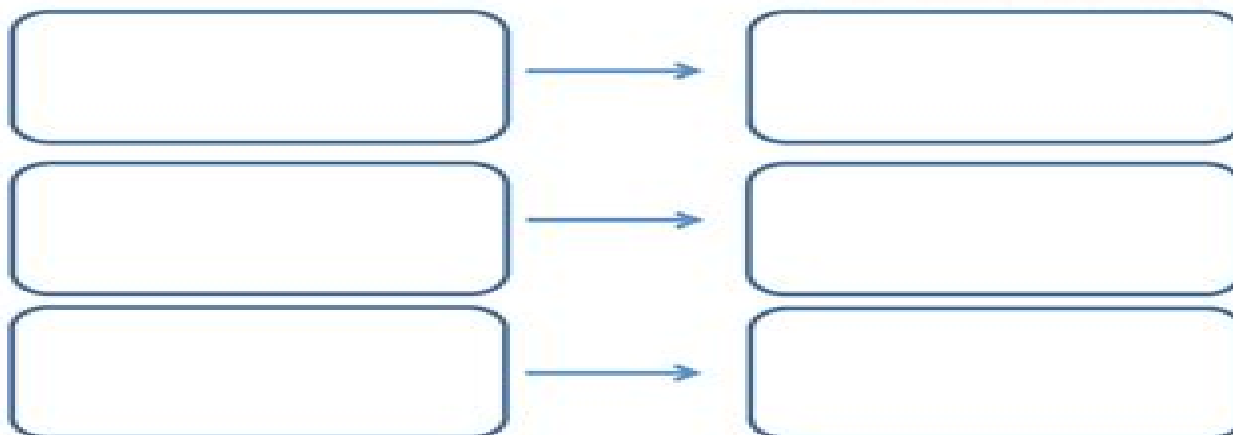


ChemThink: Chemical Reactions

In a chemical reaction, the starting material is called the _____.

The materials left in the end are called the _____.



An _____ is placed between them indicating a chemical change has taken place.

Chemical reactions always involve _____, _____, or both.

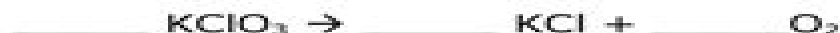
In chemical reactions, there can never be _____ or _____ when the reaction is over.

Chemical reactions only rearrange the _____ in _____ that are already there.

A reaction is _____ when it shows us not only what is present before and after the reaction but also shows _____ of each reactant and product will be present. The is called the Law of _____ of _____.

Correct.

Water formation



Chemical Reaction Chemthink Answers

Kristi Lew



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Problem-Solving Worktext James E. Brady, Frederick Senese, 2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes, both chemical and physical, that they undergo. **Fireworks** One of the ingredients is phosphorus, a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms. The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light. **Carbon** The element is found in several forms, including the familiar diamond and another recently discovered sooty substance that consists of soccer ball-shaped molecules, often referred to as buckyballs. **Diamond** is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth. **Acetylene torch** Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen, a reaction between molecules of C_2H_2 and O_2 to give CO_2 and H_2O . **Hot air balloon** The air that helps it rise is heated by the combustion of molecules of propane, each composed of three carbon and eight hydrogen atoms. **Stormy weather** The evaporation of water serves to store energy provided by the sun. Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet. *Chemical Reactions* John Joseph Griffin, 1849 *Understanding Chemistry: Chemical reactions*, 1967 *Chemical Reactions* Rachael Morlock, 2020

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