

Section 2–4 Chemical Reactions and Enzymes (pages 49–53)

TEKS FOCUS: 1A Safe lab practices; 1B Conserve resources; 2A Plan investigations; 2B Measure precisely; 2C Make inferences; 2D Communicate valid conclusions; 3A Analyze scientific explanations; 4B Cellular processes

This section describes what happens to chemical bonds during chemical reactions. It also explains how energy changes affect chemical reactions and describes the importance of enzymes.

Chemical Reactions (page 49)

1. What is a chemical reaction? It is a process that changes one set of chemicals into another set of chemicals.

2. Complete the table about chemicals in a chemical reaction.

CHEMICALS IN A CHEMICAL REACTION

Chemicals	Definition
Reactants	The elements or compounds that enter into a chemical reaction
Products	The elements or compounds produced by a chemical reaction

3. Chemical reactions always involve changes in chemical bonds.

Energy in Reactions (page 50)

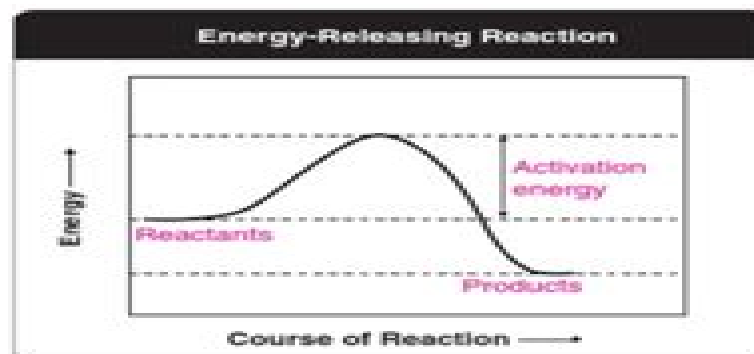
4. What is released or absorbed whenever chemical bonds form or are broken?

Energy

5. What do chemical reactions that absorb energy need to occur? They need a source of energy.

6. Chemists call the energy needed to get a reaction started the activation energy.

7. Complete the graph of an energy-releasing reaction by adding labels to show the energy of the reactants, the energy of the products, and the activation energy.



Chemical Reactions And Enzymes Workbook Answers

**Adrian Schmit, Richard Fosbery, Jenny
Wakefield-Warren**



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