

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

Double replacement/displacement

VI. Heating Cu with Atmospheric O_2

appearance the unreacted Cu mesh:

shiny copper

appearance of the Cu mesh after heating and cooling:

dull dark brown

chemical equation:



general reaction type:

combination

Which species are oxidized?

Cu

Which species are reduced?

O

VII. Reacting CuSO_4 Solution with Steel Wool (Fe)

appearance of the steel wool (Fe) before reaction:

silver solid

appearance of the CuSO_4 solution before reaction:

light blue

appearance of the steel wool (Fe) after the reaction has occurred:

dark brown solid

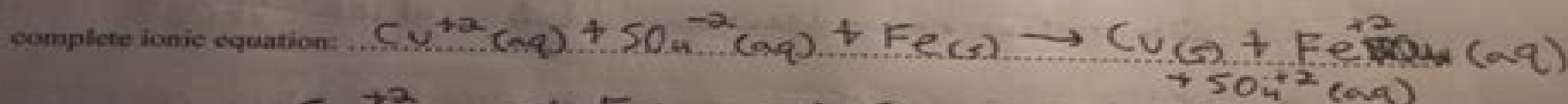
appearance of the CuSO_4 solution after the reaction has occurred:

light blue liquid

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

single displacement

Which species are oxidized?

Fe

Which species are reduced?

Cu^{+2}

Chemical Reactions And Equations Lab Answers

Valentine Korah



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