

CHEMISTRY 104

EXAM I

September 26, 2022

Directions: There are a total of 100 points for this Exam. Carefully read the directions for each section. You are free to use the provided Periodic Table and your own calculator. If a question is not clear, please do not hesitate to ask for clarification.

Useful Information: $T_C = \frac{(T_F - 32)}{1.80}$, $T_F = 1.80(T_C) + 32$, $T_K = T_C + 273$

$R = 0.082057 \text{ L-atm/mol-K}$

$1 \text{ atm} = 101.3 \text{ kPa} = 760 \text{ torr} = 760 \text{ mm Hg} = 14.7 \text{ psi}$

$PV = nRT$

$P_1V_1 = P_2V_2$

$V_1/T_1 = V_2/T_2$

$P_1/T_1 = P_2/T_2$

$P_1V_1/T_1 = P_2V_2/T_2$

$d = PM/RT$

$P_{\text{total}} = P_A + P_B + P_C + \dots$

$P_A = X_A P_{\text{total}}$

$VP_{\text{soln}} = X_{\text{solvent}} \cdot VP_{\text{pure solvent}}$

$\Delta T_b = iK_b m$

$\Delta T_f = iK_f m$

Solvent	Normal Boiling Point (°C)	K_b (°C/m)	Normal Freezing Point (°C)	K_f (°C/m)
Water, H_2O	100.0	0.51	0.0	1.86
Benzene, C_6H_6	80.1	2.53	5.5	5.12
Ethanol, $\text{C}_2\text{H}_5\text{OH}$	78.4	1.2	-114.6	.99
Carbon tetrachloride, CCl_4	76.8	5.02	-22.3	29.8
Chloroform, CHCl_3	61.2	3.6	-63.5	.68

NAME: KEY

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Table of Contents Chemistry 104 And 17 Practice Answers

1. Understanding the eBook Chemistry 104 And 17 Practice Answers
 - The Rise of Digital Reading Chemistry 104 And 17 Practice Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 104 And 17 Practice Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 104 And 17 Practice Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 104 And 17 Practice Answers
 - Personalized Recommendations
 - Chemistry 104 And 17 Practice Answers User Reviews and Ratings
 - Chemistry 104 And 17 Practice Answers and Bestseller Lists

5. Accessing Chemistry 104 And 17 Practice Answers Free and Paid eBooks
 - Chemistry 104 And 17 Practice Answers Public Domain eBooks
 - Chemistry 104 And 17 Practice Answers eBook Subscription Services
 - Chemistry 104 And 17 Practice Answers Budget-Friendly Options
6. Navigating Chemistry 104 And 17 Practice Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 104 And 17 Practice Answers Compatibility with Devices
 - Chemistry 104 And 17 Practice Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 104 And 17 Practice Answers
 - Highlighting and Note-Taking Chemistry 104 And 17 Practice Answers
 - Interactive Elements Chemistry 104 And 17 Practice Answers
8. Staying Engaged with Chemistry 104 And 17 Practice Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 104 And 17 Practice Answers
9. Balancing eBooks and Physical Books Chemistry 104 And 17 Practice Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 104 And 17 Practice Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 104 And 17 Practice Answers
 - Setting Reading Goals Chemistry 104 And 17 Practice Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 104 And 17 Practice Answers
 - Fact-Checking eBook Content of Chemistry 104 And 17 Practice Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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