Balancing Chemical Equations

Balance the following chemical equations.

I _____ N, + ____ O, → ____ N,O,

Z _____ KBr + Cl₂ → ____ KCl + Br₂

 $\boxed{3}$ $P + O_2 \rightarrow P_4O_6$

4 Al + NaOH → Na,AlO, + H,

5 Si + NaOH → Na,SiO, + H,

6 _____ Cu + ____ HNO, → ____ Cu(NO₂), + ____ NO + ___ H₂O

7 MnO₂ + Al → Mn + Al₂O₂

8 _____ HgO → ____ Hg + O₂

9 ____ KMnO₄ → ___ K₂O + ___ MnO + __ O₂

10 _____ Al + Fe,O, → ____ Fe + Al,O,

11 AgCl → Ag + Cl,

12 _____ NaCl + Ca(NO₃), → ____ NaNO₃ + CaCl,

13 SrCO, + _____ NH, Br → SrBr, + (NH,),CO,

14 C₄H₁₁ + O₇ → CO₇ + H₇O

15 _____ Ca(OH), + ____ H,PO, → Ca,(PO,), + ____ H,O

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S.K. Jain & Shailesh K. Jain

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