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Where is the hydrophilic (attracted to water) region of the molecule?

- (A) Region 1
- (B) Region 2
- (C) Region 3
- (D) The three regions are equally hydrophilic



C

Which molecule is most soluble in water?



D

A solution of NaCl in water has a concentration of 20.5% by mass. What is the molal concentration of the solution?

Molar Mass NaCl = 58.44 g/mol

- (A) 0.205 m
- (B) 0.258 m
- (C) 3.51 m (D) 4.41 m
- ER.

What is the mole fraction of water in 200 g of 89% (by mass) ethanol, C2H5OH?

Molar Mass C2H3OH = 46 g/mol

- (A) 0.11
- (B) 0.24
- (C) 0.32
- (D) 0.76

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A mixture of 100 g of K2Cr2O7 and 200 g of water is stirred at 60 degrees C until no more of the salt dissolves. The resulting solution is poured off into a separate beaker,

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Monty J. Montjar

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