

# Mark scheme

## Chemistry Past Paper Questions 1

### SECTION B: CHEMISTRY

Item	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		1
4 (c)	$(20 \times 10) + (80 \times 11)/100$ (1) = 10.8 (1)	$(200 + 80)/100$ (1) or $1080/100$ (1)	2
4 (d)	2 or 8 but not 3 for 1 mark. 2, 8, 3 for 2 marks.		2
<b>Total mark</b>			<b>6</b>

Item	Expected answers	Additional guidance	Marks
5 (a)	Any <b>one</b> of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
5 (b)	Any <b>two</b> of the following: They have similar reactions to each other (1) They have similar properties to each other (1) They all have 1 electron in their outer shell (1)	Do not accept 'they are all alkali metals'	2
<b>Total mark</b>			<b>3</b>

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The Louisiana Planter and Sugar Manufacturer, 1900      New England Journal of Education Thomas Williams Bicknell, Albert Edward Winship, Anson Wood Belding, 1879      **The Nation**, 1902      **A Dictionary of Applied Chemistry** Thomas Edward Thorpe, 1913      **Earth and Mind II** Kim A. Kastens, Cathryn A. Manduca, 2012 Articles refer to teaching at various different levels from kindergarten to graduate school with sections on teaching geologic time space complex systems

and field work Each section includes an introduction a thematic paper and commentaries  
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