

Important Question for Class 12
Chemistry
Chapter 1 – The Solid State

Very Short Answer Questions

1 Mark

1. What are anisotropic substances?

Ans: The chemical and physical properties of crystals are different in different directions. This phenomenon is known as anisotropy, and these substances are anisotropic.

2. Why are amorphous solids isotropic in nature?

Ans: Amorphous solids have an uneven particle arrangement in all directions; therefore every physical property's value is the same in any direction. As a result, they are isotropic.

3. Why glass is regarded as an amorphous solid?

Ans: In the case of glass, there is no regular, repeating molecular structure and the solid is considered to be amorphous.

4. Define the term 'crystal lattice'.

Ans: Lattice is a three-dimensional orderly arrangement of atoms, ions, and molecules. A crystal lattice is sometimes referred to as a space lattice or a lattice in its most common form.

5. Define the term voids.

Ans: Void, in its literal sense, refers to the spaces between the component particles of a substance. Unoccupied or empty space in a unit cell is what we refer to as the void.

6. What type of stoichiometric defect is shown by

(i) ZnS

Ans: There is a large difference in the size of Zn^{2+} ion and S^{2-} ion, therefore, there will be a Frenkel defect.

(ii) CsCl

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