

6. What is the molecular geometry around the central atom in SO_2 ?



7. Rank the following by increasing the boiling point



note: alcohols
hydrogen bond
to each other

8. A hydrocarbon (C_xH_y) is reacted with oxygen to produce carbon dioxide and water. What is the coefficient in front of carbon dioxide in the balanced chemical equation?

A. 10

B. 12

C. 16

D. 20

E. 24



9. Calculate the percent yield of AlCl_3 (molar mass = $133.33 \text{ g mol}^{-1}$) of a reaction that produced 14.7 g AlCl_3 from 12.5 g HCl (molar mass = $36.458 \text{ g mol}^{-1}$) in the reaction below:



A. 60.10%

B. 93.18%

C. 51.24%

D. 45.45%

E. 95.42%

10. Sucrose ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$) reacts with potassium chlorate according to the following balanced equation:



Suppose 467.5 g of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ (molar mass = $342.296 \text{ g mol}^{-1}$) are mixed with 989.5 g of KClO_4 (molar mass = $122.55 \text{ g mol}^{-1}$). What mass of H_2O (molar mass = $18.016 \text{ g mol}^{-1}$) will be formed in this reaction?

A. 145 g

B. 106 g

C. 200 g

D. 272 g

E. 224 g

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