# CHEMISTRY 1 FINAL EXAM REVIEW

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Name Section Date

#### Chemistry Review: Writing and Balancing Equations

Directions: Solve each problem and answer each question

- 1. Balance each equation below by adding the correct coefficients
  - a. \_\_SuCl. + \_Nubt -b \_NuCl + \_Subt.,
  - St. \_\_HCN + \_\_Mg(DH)\_ (t \_\_Mg(DN)\_ + \_\_HUD
  - E. \_\_HWO,+\_\_IMH,3,50,+\_\_MH,MO,+\_\_H,50,
  - 4. \_2MDDJ+\_ALO, +\_2HO+\_ABDDJ,
  - 4. \_CH.+\_0,+\_00,+\_N0
- 2: Write and bulance each equation, Include phase labels
  - Aqueous sodium hydroxide reacts with aqueous suffuric acid to form aqueous sodium suffute and solver
  - Solid iron (III) hydroxide and approxis sollium carbonate are formed from the reaction of solid iron (III) carbonate and appears another hydroxide
  - Solid aluminum reacts with agreens hydrotromic acid to form agreess aluminum bromide and hydrogen gas
  - Gaseous ethane (dicarbon heaphydride) reacts with saygengas to form gaseous carbon disable and gaseous water
  - Solid calclum phosphite reacts with asserous opport(ii) retrate to form asserous calcium retrate and solid copper (II) phosphite

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John T. Moore, Richard H. Langley

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