


Chemistry Semester 1 Final Exam Review

1. What is a definition of Chemistry?
Study of matter
 2. Matter includes all of the following EXCEPT:
a. air **b. light** c. smoke d. water vapor
 3. What is matter?
Something that has mass and takes up space (volume)
 4. A physical change occurs when a
a. peach spoils c. copper bowl tarnishes
b. bracelet turns your wrist green **d. glue gun melts a glue stick**
 5. The particles in a solid are
a. packed closely together c. constantly in motion
b. very far apart d. able to slide past one another
- Of the following groups, 1, 2, 14, 15, 17, 18:
6. Group(s) **1, 2** contains only metals
 7. Group (s) **18** is called a Noble gases
 8. Group(s) **17** has 7 electrons on the valence shell
 9. Group(s) **15** needs to gain 3 electrons to fill a stable octet
 10. Group(s) **2** has a 2+ oxidation state
 11. Group(s) **2** has an electron dot notation of $\bullet X \bullet$
 12. Using the periodic table, an element similar to carbon would be **14**
 13. Based on their location on the periodic table, you could infer that **18/noble gases** are very unreactive
 14. The horizontal row on the periodic table is called a(n) **period**
 15. The vertical column of blocks on the periodic table is called a(n) **group/family**
 16. The elements on the zigzag line in the periodic table are **metalloids**
 17. Which is NOT a property of metal?
a. malleability c. **unreactivity**
b. ability to lose electrons d. ability to conduct heat
 18. A volume of 1 cc (cm^3) is the same as **1** mL
 19. To determine density, the quantities that must be measured are **volume** and **mass**
 20. The relationship between the mass m of a material, its volume V , and its density D is
a. $V = mD$ b. $V/m = D$ c. **$DV = m$** d. $D + V = m$
 28. The density of aluminum is 2.70 g/cm^3 . The volume of a solid piece of Aluminum is 1.50 cm^3 . Find the mass **4.05 grams**
 21. The density of pure diamond is 3.5 g/cm^3 . The mass of a diamond is 0.25 g. Find the volume= **0.71 cm^3**
 22. What statement about density is true?
a. two samples of a pure substance may have different densities if they are different sizes
b. the density of a sample depends on its location on Earth
c. a cylinder is always used to measure the volume
d. **density is a physical property that remains constant for a pure substance**
 23. The number of grams equal to 0.5 kg is **500 g**
 24. In a graph the **independent** variable is placed on the x axis and the **dependant** variable is placed on the y axis.
 25. The variable that you change in the lab is the **independent** variable. The variable that responds to the change is called the **dependant** variable. All other things do not change and are called **constants**
 26. At first Dalton thought the atom was solid, spherical, **indivisible**, then Rutherford discovered the **nucleus** of the atom, then the subatomic particles **protons**, **neutrons**, and **electrons** were discovered
 27. The positively charged particle is the **proton** found in the **nucleus** of the atom
 28. The subatomic particle that has about the same mass as the proton, but with no electrical charge, is called a **neutron** and is found in the **nucleus** of the atom
 29. The **electron** is negatively charged and is found in the **electron cloud/outer shell** of the atom
 30. **Bohr** is the person credited with placing electrons on levels and **Schrodinger** established the electron cloud theory.
 31. The forces that hold the particles together in the nucleus are **columbic /electrostatic** forces
 32. Isotopes are atoms of the same element that **have different number of neutrons**
 33. All isotopes of hydrogen contain **1 proton & 1 electron**
 34. What is the atomic number for Nitrogen? **7** To what subatomic particle does this correspond? **# of protons**
 35. Zinc -66 has **30** protons and **36** neutrons
 36. What is the electron configuration for sodium? **$[\text{Ne}]3s^1$** Chlorine? **$[\text{Ne}]3s^23p^5$** Neon? **$[\text{He}]2s^22p^6$**
 37. What is the dot notation for magnesium? Bromine? Argon?

Chemistry 1 Semester Exam Review Answers

Richard H. Langley



Chemistry 1 Semester Exam Review Answers:

Nursing School Entrance Exam Review William H. Brown, Frederick A. Bettelheim, Jerry March, 2004-04 **Artificial Intelligence in Education. Posters and Late Breaking Results, Workshops and Tutorials, Industry and Innovation Tracks, Practitioners, Doctoral Consortium and Blue Sky** Andrew M. Olney, Irene-Angelica Chounta, Zitao Liu, Olga C. Santos, Ig Ibert Bittencourt, 2024-07-01 This volume constitutes poster papers and late breaking results presented during the 25th International Conference on Artificial Intelligence in Education AIED 2024 which took place in Recife Brazil during July 8 12 2024 The 18 full papers and 92 short papers were carefully reviewed and selected from 200 submissions They are organized in topical sections as follows Part One Blue Sky Industry Innovation and Practitioner Wide AIED and Late Breaking Results Part Two Late Breaking Results Doctoral Consortium Workshops and Tutorials **The Software Encyclopedia 2001** , 2001 Cultural Issues: Creation/Evolution and the Bible (Teacher Guide) Ken Ham, 2016-09-06 The vital resource for grading all assignments from the Cultural Issues Creation Evolution and the Bible course which includes Learning answers information and strategies when facing destructive influences found in the workplace or school environments Studying fossils the age of the earth the beginning of life and more in these two volumes focused on points of contention related to the Bible faith and science OVERVIEW This curriculum has been put together to provide the answers to many common objections to biblical worldviews and scriptural authority of the Bible Practical tests are included to strengthen the student s grasp of key concepts and terms while providing critical thinking opportunities to put their knowledge to work Students will learn to apply the Biblical worldview to subjects such as evolution carbon dating Noah s ark and the Flood and dozens more They will discover answers to help know the depths of God s wisdom found in His Word and in His world and why this matters to your life your family and your faith FEATURES The calendar provides lesson planning with clear objectives and the worksheets and tests are all based on the materials provided for the course *Resources in Education* , 1990-05 **5 Steps to a 5 AP Chemistry** John Moore, 2003-08-22 For the more than one million students taking the AP exams each year Boxed quotes offering advice from students who have aced the exams and from AP teachers and college professors Sample tests that closely simulate real exams Review material based on the contents of the most recent tests Icons highlighting important facts vocabulary and frequently asked questions Websites and links to valuable online test resources along with author e mail addresses for students with follow up questions Authors who are either AP course instructors or exam developers *Listening to Our Students and Transcending K-12 to Save Our Nation* Alec Ostrom, Brian Hack, Don Prentice, 2020-01-06 This journey will engage you in dealing with some hard truths and it will take you down a new pathway and new ways of thinking about K 12 education We now live in a nation that is struggling with deep social economic and political conflicts We are all doing our best to resolve these conflicts and to solve the critical challenges that we all face in the Digital Age but our children and young adults are having a very difficult time in dealing with the realities of their

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Table of Contents Chemistry 1 Semester Exam Review Answers

1. Understanding the eBook Chemistry 1 Semester Exam Review Answers
 - The Rise of Digital Reading Chemistry 1 Semester Exam Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1 Semester Exam Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1 Semester Exam Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 1 Semester Exam Review Answers
 - Personalized Recommendations
 - Chemistry 1 Semester Exam Review Answers User Reviews and Ratings
 - Chemistry 1 Semester Exam Review Answers and Bestseller Lists
5. Accessing Chemistry 1 Semester Exam Review Answers Free and Paid eBooks
 - Chemistry 1 Semester Exam Review Answers Public Domain eBooks
 - Chemistry 1 Semester Exam Review Answers eBook Subscription Services
 - Chemistry 1 Semester Exam Review Answers Budget-Friendly Options
6. Navigating Chemistry 1 Semester Exam Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 1 Semester Exam Review Answers Compatibility with Devices
 - Chemistry 1 Semester Exam Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 1 Semester Exam Review Answers
 - Highlighting and Note-Taking Chemistry 1 Semester Exam Review Answers
 - Interactive Elements Chemistry 1 Semester Exam Review Answers
8. Staying Engaged with Chemistry 1 Semester Exam Review Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry 1 Semester Exam Review Answers
- 9. Balancing eBooks and Physical Books Chemistry 1 Semester Exam Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1 Semester Exam Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 1 Semester Exam Review Answers
 - Setting Reading Goals Chemistry 1 Semester Exam Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 1 Semester Exam Review Answers
 - Fact-Checking eBook Content of Chemistry 1 Semester Exam Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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