

CHEMISTRY 11 – WORD EQUATIONS AND TYPES OF CHEMICAL REACTIONS WORKSHEET

- 1) Translate the following word equations into chemical symbols and balance the resulting equations. Do not include the phases.

(a) Potassium + water → potassium hydroxide + hydrogen



(b) Nitrogen + hydrogen → ammonia



- 2) In each of the following write a balanced chemical equation, including the phases.

(a) Liquid phosphoric acid reacts with aqueous barium hydroxide to give water and a precipitate of barium phosphate



(b) Solid aluminium oxide and aqueous sulphuric acid produce water and aqueous aluminium sulphate



(c) Nitrogen trifluoride gas and hydrogen gas react to form nitrogen gas and gaseous hydrogen fluoride



(d) Powdered sodium carbonate and aqueous hydrobromic acid react to form carbon dioxide gas, aqueous sodium bromide and water



- 3) Classify the following reactions.

Reaction	Type
$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$	Combustion
$\text{Mg} + \text{CuSO}_4 \rightarrow \text{MgSO}_4 + \text{Cu}$	Single Replacement
$4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$	Synthesis
$\text{Sn}(\text{OH})_4 + 4\text{HBr} \rightarrow 4\text{H}_2\text{O} + \text{SnBr}_4$	Neutralization
$2\text{Fe}(\text{NO}_3)_3 + 3\text{MgS} \rightarrow \text{Fe}_2\text{S}_3 + 3\text{Mg}(\text{NO}_3)_2$	Double Replacement
$2\text{N}_2\text{O} \rightarrow 2\text{N}_2 + \text{O}_2$	Decomposition

Chemistry 1f8766 Word Equations

George R. Hague, Jane D. Smith



Chemistry 1f8766 Word Equations:

Chemical Equations Lifeliqe,2019 This lesson plan covers describing chemical reactions using word equations knowing the correct symbols to use to write skeleton equations for chemical reactions and balancing chemical equations

Coordinating Procedural and Conceptual Knowledge to Make Sense of Word Equations: Understanding the Complexity of a 'simple' Chemical Task at the Learner's Resolution Keith Stephen Taber,Pat Bricheno,2009 Abstract This paper discusses the conceptual demands of an apparently straightforward task set to secondary level students completing chemical word equations with a single omitted term Chemical equations are of considerable importance in chemistry and school students are expected to learn to be able to write and interpret them However it is recognized that many students find them challenging The present paper explores students accounts of their attempts to identify the missing terms to illuminate why working with chemical word equations is so challenging from the learner s perspective 300 secondary age students responded to a 5 item exercise based on chemicals and types of reactions commonly met at school level For each item they were asked to identify the missing term in a word equation and explain their answers This provided a database containing more than a thousand student accounts of their rationales Analysis of the data led to the identification of seven main classes of

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