

Calcium

- Found as a salt in bones and teeth
- In ionic form, required for muscle contraction, neural transmission, and blood clotting.

Phosphorus

- Present as salt, in combination with calcium in bones and teeth
- Present in nucleic acids and many proteins
- Forms part of the higher energy compound ATP

Potassium

- In its ionic form, the major intracellular cation
- Necessary for the conduction of nerve impulses and for muscle contraction.

Sulfur

- A component of proteins (particularly contractile proteins of muscle)

Element	Percentage of body mass
Sodium (Na)	0.2
Chlorine (Cl)	0.2
Magnesium (Mg)	0.1
Iodine (I)	0.1
Iron (Fe)	0.1

Sodium

- As an ion, the major extracellular cation
- Important for water balance, conduction of nerve impulses and muscle contraction

Chlorine

- In ionic (Chloride) form, the most abundant extracellular anion

Magnesium

- Present in bone
- Also important cofactor for enzyme activity in a number of metabolic reactions

Iodine

- Needed to make functional thyroid hormones

Iron

- A component of the functional hemoglobin molecule (which transports oxygen within red blood cells) and some enzymes

Atoms

- Building blocks of elements
- Atoms of elements differ from one another
- Atomic symbol is chemical shorthand for each element

The Basic Atomic Sub particles

- **Protons (p^+)** are positively charged
- **Neutrons (n^0)** are **uncharged or neutral**
- **Electrons (e^-)** are negatively charged

All atoms are electrically neutral

- Number of protons equals numbers of electrons in an atom
- Positive and negative charges cancel each other out

Ions are atoms that have lost or gained electrons

Planetary and Orbital Models of an Atom**Planetary model**

- Portrays the atom as a miniature solar system
- Protons and neutrons are in the atomic nucleus
- Electrons are in orbitals around the nucleus

Orbital model

- Electrons are depicted by an electron cloud, a haze of negative charge, outside the nucleus

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Joacim Rocklöv



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