

Chapter 2

I. Atoms and Molecules

- A. Emergent properties – atoms and molecules
- B. Life requires about 25 chemical elements
- C. Elements combine to form compounds - 2 or more elements in fixed ratio
 - 1. table salt is NaCl; iodized (fig. 2.4)
 - 2. Water
- D. Atoms consist of protons, neutrons and electrons
 - 1. atom is the smallest unit of matter
 - 2. subatomic particles make up the atom.
 - 3. Differences in elements
 - 4. isotopes - different numbers of neutrons so mass changes
- E. Electron arrangement determines chemical properties of atom
 - 1. electrons determine how an atom behaves
 - 2. electrons vary in energy
 - 3. electrons occur at certain energy levels
- F. Ionic bonds are attractions between ions of opposite charges
 - 1. ion = loss or gain of electron
 - 2. ions with opposite charges attract each other
 - 3. resulting compound is electrically neutral
 - 4. sodium and chloride ions are always present in 1:1 ratio
- G. Covalent bonds = sharing of electrons in outer shell
 - 1. molecule
 - 2. single covalent bond
 - 4. double bond
 - 5. H₂ and O₂ are molecules but not compounds
- H. Chemical reactions rearrange matter
 - 1. chemical reactions
 - 2. reactants
 - 3. product
 - 4. $2\text{H}_2 + \text{O}_2 = \text{H}_2\text{O}$

II. Properties of water

- A. water is a polar molecule
- B. Water's polarity leads to H bonding
- C. Hydrogen bonds make liquid water cohesive
 - 1. H bonds last for only a few trillionths of a second
 - 2. cohesion and adhesion
 - 3. surface tension
- D. Hydrogen bonds of water moderate temperature
 - 1. heat
 - 2. temperature
- E. Ice is less dense than water
- F. water is a versatile solvent
 - 1. solution
 - 2. solvent
 - 3. solute
 - 4. aqueous solution

III. pH Scale

- A. The chemistry of life is sensitive to acid and basic conditions
 - 1. some water molecules break apart naturally in water
 - 2. ions formed are called hydrogen (H⁺) and hydroxide (OH⁻) ions
 - 3. for life chemistry, proper balance of these ions is necessary
 - 4. the more acidic a solution, the more H⁺ ions. Acids are substances that donate H⁺
 - 5. in your stomach, HCl donates H⁺ ions when in solution

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Mark Hawley, 2017-04 Guidelines for Mine Waste Dump and Stockpile Design is a comprehensive practical guide to the investigation design operation and monitoring of mine waste dumps dragline spoils and major stockpiles associated with large open pit mines These facilities are some of the largest man made structures on Earth and while most have performed very well there are cases where instabilities have occurred with severe consequences including loss of life and extensive environmental and economic damage Developed and written by industry experts with extensive knowledge and experience this book is an initiative of the Large Open Pit LOP Project It comprises 16 chapters that follow the life cycle of a mine waste dump dragline spoil or stockpile from site selection to closure and reclamation It describes the investigation and design process introduces a comprehensive stability rating and hazard classification system provides guidance on acceptability criteria and sets out the key elements of stability and runout analysis Chapters on site and material characterisation surface water and groundwater characterisation and management risk assessment operations and monitoring management of ARD emerging technologies and closure are included A chapter is also dedicated to the analysis and design of dragline spoils Guidelines for Mine Waste Dump and Stockpile Design summarises the current state of practice and provides insight and guidance to mine operators geotechnical engineers mining engineers hydrogeologists geologists and other individuals that are responsible at the mine site level for ensuring the stability and performance of these structures Readership includes mining engineers geotechnical engineers civil engineers engineering geologists hydrogeologists environmental scientists and other professionals involved in the site selection investigation design permitting construction operation monitoring closure and reclamation of mine waste dumps and stockpiles *Books in Print Supplement*, 2002 **Index to Media and Materials for the Mentally Retarded, Specific Learning Disabled, Emotionally Disturbed** National Information Center for Special Education Materials, 1978 **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter **Organic Chemistry Laboratory** Charles E. Bell, 1997 *The Software Encyclopedia 2001*, 2001

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Table of Contents Chapter 2 Basic Chemistry Worksheet Answers

1. Understanding the eBook Chapter 2 Basic Chemistry Worksheet Answers
 - The Rise of Digital Reading Chapter 2 Basic Chemistry Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 2 Basic Chemistry Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 2 Basic Chemistry Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 2 Basic Chemistry Worksheet Answers
 - Personalized Recommendations
 - Chapter 2 Basic Chemistry Worksheet Answers User Reviews and Ratings
 - Chapter 2 Basic Chemistry Worksheet Answers and Bestseller Lists
5. Accessing Chapter 2 Basic Chemistry Worksheet Answers Free and Paid eBooks
 - Chapter 2 Basic Chemistry Worksheet Answers Public Domain eBooks
 - Chapter 2 Basic Chemistry Worksheet Answers eBook Subscription Services
 - Chapter 2 Basic Chemistry Worksheet Answers Budget-Friendly Options

6. Navigating Chapter 2 Basic Chemistry Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 2 Basic Chemistry Worksheet Answers Compatibility with Devices
 - Chapter 2 Basic Chemistry Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 2 Basic Chemistry Worksheet Answers
 - Highlighting and Note-Taking Chapter 2 Basic Chemistry Worksheet Answers
 - Interactive Elements Chapter 2 Basic Chemistry Worksheet Answers
8. Staying Engaged with Chapter 2 Basic Chemistry Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 2 Basic Chemistry Worksheet Answers
9. Balancing eBooks and Physical Books Chapter 2 Basic Chemistry Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 2 Basic Chemistry Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 2 Basic Chemistry Worksheet Answers
 - Setting Reading Goals Chapter 2 Basic Chemistry Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 2 Basic Chemistry Worksheet Answers
 - Fact-Checking eBook Content of Chapter 2 Basic Chemistry Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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