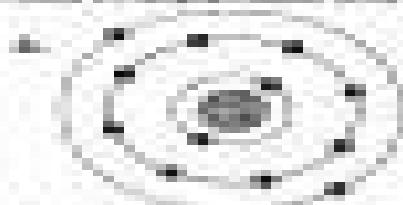


IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet

Directions: Study the diagrams below, then give a brief description of each type of bond.



3. Fill in the blanks (use information from 1).

- How many electrons will atom A gain? _____
- How many electrons will atom B gain? _____
- What will be the charge of atom A? _____
- What will be the charge of atom B? _____
- What will be the bond charge of the compound formed? _____
- What type of bond will form? _____

4. Explain why a covalent bond is stronger than ionic bonds (use the periodic table to check charges).

Directions: Complete the table comparing the properties of ionic and covalent compounds.

Compounds	Ionic compounds	Covalent compounds
How they compound is formed	_____	_____
Strongest particle	_____	_____
State of matter at room temperature	_____	_____

Chapter 20 Section 2 Types Of Bonds Worksheet Answers

Jean Craven



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