

Section 21.5 Aromatic Hydrocarbons

In your textbook, read about the structure of benzene and aromatic compounds.

Use the following words to complete the statements.





aliphatic compounds

aromatic compounds

carcinogens

1. Substances that cause cancer are called _____.
2. Alkanes, alkenes, and alkynes are examples of _____.
3. All _____ contain benzene rings as part of their structure.

Circle the letter of the choice that best completes the statement or answers the question.

4. What is the molecular formula of benzene?
a. C_6H_6 b. C_6H_{12} c. C_6H_{14} d. $C_{12}H_{12}$
5. Which of the following is the best way to represent the structure of benzene?
a.  b.  c.  d. 
6. Compared to alkenes and alkynes of similar size, benzene is
a. less reactive. c. slightly more reactive.
b. about as reactive. d. much more reactive.
7. The electrons in the ring of an aromatic compound are
a. held tightly by one carbon nucleus.
b. localized between specific carbon nuclei.
c. shared equally by all of the carbon nuclei.
d. shared by only three of the carbon nuclei.
8. The use of aromatic compounds should be limited because many of them
a. produce chimney soot. c. have pleasant smells.
b. can cause health problems. d. cannot be synthesized.

Use the IUPAC rules to name the following structures.





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Robert R. Redfield



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