



Chemical Changes

Answer the questions below to check the following questions:



1. What are the chemical symbols for hydrogen and oxygen?

H and O₂

2. What is the product? (2)

3. What is the physical state of the products and the reactants?

The products are liquids and the reactants are gases.

4. According to the law of conservation of mass, if 100g of water and 10g of oxygen are placed in a bag, what mass of the products will be formed?

110g



5. What products are formed? (2)

6. What mass of the products is formed? (2)

7. What are the physical states of the products and the reactants?

Liquids and gases

8. What is the physical state of the product? (2)

9. What is the physical state of the reactants? (2)

They are gases

10. What mass of the product is formed? (2)

Answer the following questions for the following reaction:

11. How many molecules of H_2O are produced if 100g of H_2 and 100g of O_2 are placed in a bag?

100g of H_2O is produced

Chapter 24 Section Reinforcement Chemical Changes

**Sharon L. Lewis, Shannon Ruff
Dirksen, Margaret M. Heitkemper, Linda
Bucher, Ian Camera**



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