

# Chemical Reactions

## Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- |               |   |
|---------------|---|
| <u>yes</u>    | 1. A tomato smells rotten.  |
| <u>no</u>     | 2. A drinking glass breaks into smaller pieces.                     |
| <u>no</u>     | 3. A piece of ice melts.  |
| <u>yes</u>    | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u>    | 5. Candle wax burns.  |
| <u>yes/no</u> | 6. Molten candle wax solidifies.                                    |
| <u>yes</u>    | 7. Green leaves turn yellow and red as the seasons change.          |
| <u>yes</u>    | 8. Baking powder produces a gas that makes a cake rise.             |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

# Chapter 1 chemical Reactions Vocabulary Review Answer Key

**J Ma**



## **Chapter 1 chemical Reactions Vocabulary Review Answer Key:**

**Chemistry (Teacher Guide)** Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study.

**Features** Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule.

**Workflow** Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

**About the Author** DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

*Focus on physical science* Charles H. Heimler, 1989

**Prehospital Behavioral Emergencies and Crisis Response** American Academy of Orthopaedic Surgeons (AAOS),, Dwight A. Polk, Jeffrey T. Mitchell, 2008-12-09 Listen to a podcast with author Dwight Polk. Prehospital Behavioral Emergencies and Crisis Response offers prehospital providers and mental health professionals a comprehensive resource on effectively handling behavioral emergencies ranging from large scale disasters to substance related emergencies to psychotic disorders. This timely text also provides prehospital providers with suggestions for coping with job related stress. The text incorporates well known information gathering tools such as SAMPLE and OPQRST as well as two essential new tools for evaluating patients in a behavioral emergency: the SEA 3, a brief and simple assessment tool designed specifically for gathering information about a patient's mental status, and the SAFER R

model a simple yet thorough tool designed to help the emergency responder devise a quick and easy treatment plan for helping those in crisis In the Field boxes are scattered throughout the chapters and highlight vital information tips and best practices that first responders can apply during an emergency call End of chapter material provides vocabulary chapter summaries and quizzes that allow the reader to check his or her knowledge and comprehension of the chapter Important tips for applying concepts to real life situations first responders encounter in the field including step by step recommendations for conducting assessments Quiz questions to evaluate students understanding of the material Crucial vocabulary that first responders should know when responding to behavioral emergencies Easy to follow methods for quickly evaluating behavioral emergencies using familiar information gathering tools Part of our Continuing Education Series instructors can use the manual and instructor resources for specific classes on behavioral emergencies or as a supplement to existing EMT B EMT I or paramedic classes 2009 298 pages *Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed* ,2000 **The Practice of Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB *Electricity and Magnetism* ,2005 *Scott Foresman-Addison Wesley Middle School Math* ,2002 **Modules** McDougal Littell Incorporated, 2005 *Glencoe Science* McGraw-Hill Staff, 2001-06 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Science Insights** ,1999 **Addison-Wesley Science Insights** ,1996 **Student Edition** Glencoe, 2001-05 *Harcourt Science: Life science, units A and B* ,2002 **Physical Science Anted, Grades 6-9** David W. Appenbrink, 1986 **Chemical Interactions** ,2005 *Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed* ,2000 *Teen Health* Mary H. Bronson, Michael J. Cleary, Betty M. Hubbard, Dinah Zike, 2005 **Addison-Wesley Health and Safety** Stuart Lazarus, 1989 *Biology* Joseph S. Levine, Kenneth R. Miller, 1998

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