

Behavior of Gases

KEY IDEAS

As you read this section, keep these questions in mind:

- What are some properties of gases?
- How do changes of pressure, temperature, or volume affect a gas?

What Are Some Properties of Gases?

Particles in a gas move rapidly in all directions. Some of the unique properties of gases are listed below:

- expand to fill their containers
- easily mix with one another
- have low densities
- can be compressed
- are mostly empty space

GASES AND THEIR CONTAINERS

Gases exert pressure on their containers. For example, as helium molecules inside a balloon move, they bump into each other and the walls of the balloon. One molecule alone does not have a large effect. However, millions of molecules create a steady force. If too many gas molecules are in the balloon, the total pressure they exert can cause the balloon to break.



Gas particles exert pressure on the walls of a balloon.

A gas under pressure will escape its container if possible. For example, if you open the end of a balloon, gas will rush out of the balloon. For this reason, gases in pressurized containers, such as propane tanks or helium tanks, can be very dangerous.

READING TOOLBOX

Organize As you read, make a table that lists all of the gas laws discussed in the section. For each law, identify which factor must stay constant, which variables change, and the relationship between the variables.

LOOKING CLOSER

1. **Predict** What would happen to pressure if you removed some of the gas particles from the balloon?

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