

**CHAPTER 11 REVIEW***Molecular Composition of Gases***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ The average speed of a gas molecule is most directly related to the \_\_\_\_\_.
  - (a) polarity of the molecule
  - (b) pressure of the gas
  - (c) temperature of the gas
  - (d) number of moles in the sample
2. \_\_\_\_\_ Which of the following statements best represents Graham's law for two gases at the same temperature?
  - (a) The molar masses and the velocities are inversely proportional.
  - (b) The molar masses and the velocities are directly proportional.
  - (c) The molar masses and the square roots of the velocities are directly proportional.
  - (d) The molar masses and the squares of the velocities are inversely proportional.
3. Water is synthesized from its elements according to the following equation:
$$2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l})$$
  - \_\_\_\_\_ a. Which is greater, the volume of  $\text{H}_2$  consumed or the volume of  $\text{O}_2$  consumed?
  - \_\_\_\_\_ b. Which is greater, the mass of  $\text{H}_2$  consumed or the mass of  $\text{O}_2$  consumed?
4. Consider a 2.24 L sample of an ideal gas at STP.
  - \_\_\_\_\_ a. How many moles are present in the sample, or is more data needed?
  - \_\_\_\_\_ b. How many molecules are present in the sample, or is more data needed?
  - \_\_\_\_\_ c. What is the mass of the sample, or is more data needed?

**PROBLEMS** Write the answer on the line to the left. Show all your work in the space provided.

5. \_\_\_\_\_ Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

# Chapter 1review Molecular Composition Of Gases Mixed Answers

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