

## Study Guide-Stoichiometry

1. By definition the \_\_\_\_\_ of an element tells how many protons there are in the element's nucleus.
2. The atomic number determines what type of atom an element is and no two atoms have the same number of \_\_\_\_\_.
3. Since all atoms are electrically \_\_\_\_\_ neutral then they must have the same number of positive protons as negative electrons, thus the atomic number also tells you how many electrons an atom contains.
4. The \_\_\_\_\_ of an atom is equal to the sum of its protons and

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**Robinson**



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