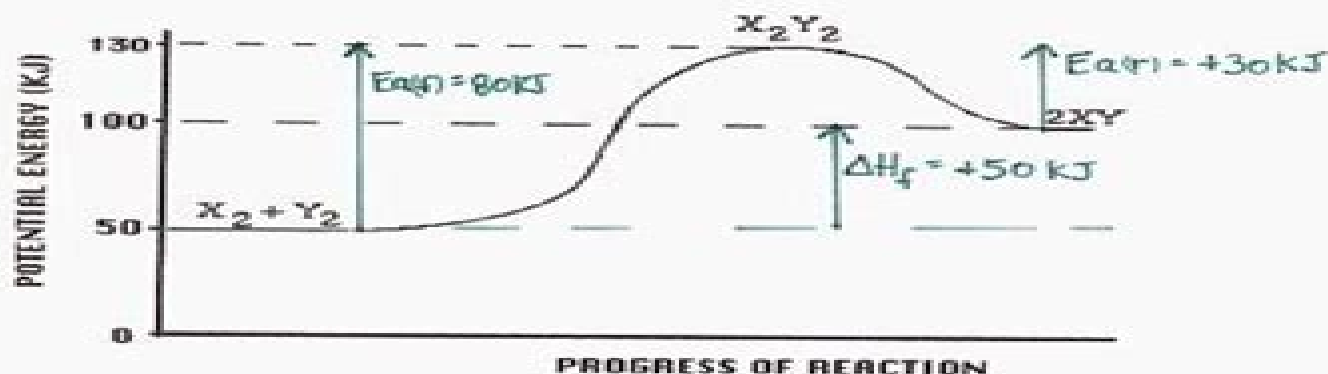


Chemistry 12
Worksheet 1-2 - Potential Energy Diagrams

KEY

USE THE POTENTIAL ENERGY DIAGRAM TO ANSWER THE QUESTIONS BELOW:



- Is the overall reaction as shown **exothermic** or **endothermic**? endothermic
- What is the **activation energy** for the forward reaction? +80 kJ
- What is the **activation energy** for the reverse reaction? +30 kJ
- What is the **enthalpy change of reaction** (ΔH) for the *forward* reaction? +50 kJ
- What is the ΔH for the reverse reaction? -50 kJ
- Is the *reverse* reaction **exothermic** or **endothermic**? exothermic
- Which species forms the *activated complex*? X_2Y_2
- Which species or set of species has the *highest potential energy*? X_2Y_2

Chapter 1 wordwise Energy Answers

Daniela Niemeyer



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