

# Chemistry of Quinoline & Isoquinoline

## Session Objectives

By the end of this session, students will be able to:

- Discuss various methods of synthesis of Quinoline and Isoquinoline
- Discuss the chemistry, reactivity, properties of Quinoline and Isoquinoline

## Chemistry of Quinoline

- Quinoline structure obtained by ortho-condensation of benzene ring with pyridine
- Also called as azanaphthalene or benzopyridine
- First isolated by Runge in 1834, from coal tar bases

• In 1842 by Gerhardt from the alkaline pyrolysis of cinchonine, an alkaloid related to quinoline

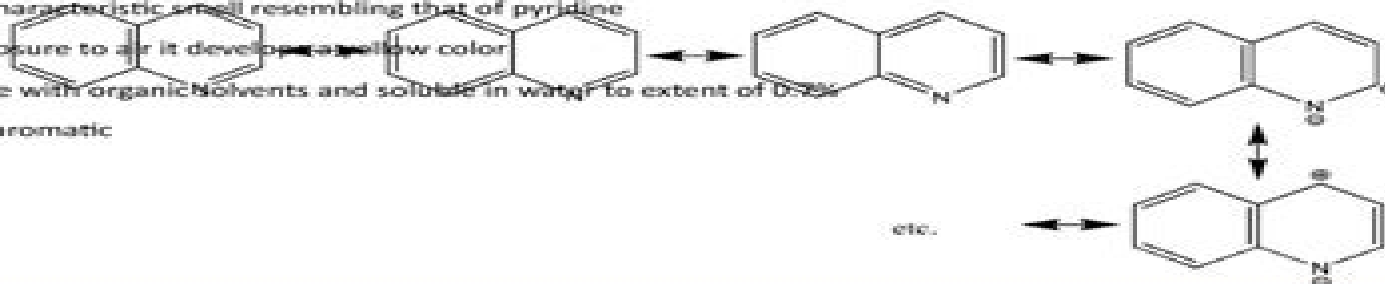
• Numbering starts from nitrogen atom, which is assigned position-1

- Dehydrogenation of quinoline yields 1,2-dihydroquinoline (1), 1,2,3,4-tetrahydroquinoline (2) and decahydroquinoline (3)



## Physical properties of Quinoline:

- Quinoline is a colorless hygroscopic liquid with boiling point 237°C
- Has a characteristic smell resembling that of pyridine
- On exposure to air it develops a yellow color
- Miscible with organic solvents and soluble in water to extent of 0.7%
- Highly aromatic



# Chapter 25 Chemistry Test Core Teaching Resources

**Huangqi Zhang**



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## **Table of Contents Chapter 25 Chemistry Test Core Teaching Resources**

1. Understanding the eBook Chapter 25 Chemistry Test Core Teaching Resources
  - The Rise of Digital Reading Chapter 25 Chemistry Test Core Teaching Resources
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 25 Chemistry Test Core Teaching Resources
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 25 Chemistry Test Core Teaching Resources
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 25 Chemistry Test Core Teaching Resources
  - Personalized Recommendations
  - Chapter 25 Chemistry Test Core Teaching Resources User Reviews and Ratings
  - Chapter 25 Chemistry Test Core Teaching Resources and Bestseller Lists
5. Accessing Chapter 25 Chemistry Test Core Teaching Resources Free and Paid eBooks
  - Chapter 25 Chemistry Test Core Teaching Resources Public Domain eBooks
  - Chapter 25 Chemistry Test Core Teaching Resources eBook Subscription Services
  - Chapter 25 Chemistry Test Core Teaching Resources Budget-Friendly Options
6. Navigating Chapter 25 Chemistry Test Core Teaching Resources eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 25 Chemistry Test Core Teaching Resources Compatibility with Devices
  - Chapter 25 Chemistry Test Core Teaching Resources Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 25 Chemistry Test Core Teaching Resources
  - Highlighting and Note-Taking Chapter 25 Chemistry Test Core Teaching Resources
  - Interactive Elements Chapter 25 Chemistry Test Core Teaching Resources
8. Staying Engaged with Chapter 25 Chemistry Test Core Teaching Resources

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chapter 25 Chemistry Test Core Teaching Resources
- 9. Balancing eBooks and Physical Books Chapter 25 Chemistry Test Core Teaching Resources
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 25 Chemistry Test Core Teaching Resources
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 25 Chemistry Test Core Teaching Resources
  - Setting Reading Goals Chapter 25 Chemistry Test Core Teaching Resources
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 25 Chemistry Test Core Teaching Resources
  - Fact-Checking eBook Content of Chapter 25 Chemistry Test Core Teaching Resources
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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