



- a. A
b. B

- c. C
d. D

10. The energy as heat produced by a reactor is used to
 a. boil water for steam turbines.
 b. melt metal.
 c. produce graphite.
 d. produce coal.
11. At present, fusion reactions
 a. cannot be used to produce energy in reactors.
 b. produce the energy in some nuclear power plants.
 c. produce the energy in most nuclear power plants.
 d. produce the energy in all recent nuclear power plants.
12. What is the product of β -ray emission from a radioactive isotope of lead?
 a. Thallium
 b. Mercury
 c. Bismuth
 d. Lead
13. What is the reason for the decay of naturally radioactive elements?
 a. To reduce the number of neutrons as the elements lie above the band of stability.
 b. To reduce the number of protons as the elements lie below the band of stability.
 c. To reduce the number of either neutrons or protons to lie in the band of stability.
 d. To release the energy from nucleus in the form of gamma radiation.
14. How does a positron differ from an electron?
 a. A positron has no electrical charge, unlike an electron.
 b. A positron has a charge opposite that of an electron.
 c. A positron has less mass than an electron.
 d. A positron has greater mass than an electron.
15. What is the main product of the fusion reactions in our sun?

Chapter 25 Nuclear Chemistry Section Review Answer Key Workbook

William L. Ramsey



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