Answer Key States of Matter

Lesson 1

Before You Read

- Disagree
- Disagree

Read to Learn

- solid, liquid, gas, plasma
- As the motion of particles gets faster, particles move farther apart.
- 3. Students should circle the third picture from the left.
- Possible answer: I would fall off the skateboard because it would not have a definite shape.
- Solid particles vibrate in place; they have strong attractive forces between the particles.
- The particles in an amorphous solid are randomly arranged in fixed positions. The particles in a crystalline solid are arranged in a fixed but repeating pattern.
- Particles in a liquid can move and slip past other particles because the attractive forces between the particles are not as strong as they are in a solid.
- Possible answer: An insect could not walk on rubbing alcohol because it has a lower surface tension than water.
- the downward pull on the water molecules at the surface of water
- Gas is a state of matter that has no definite volume and no definite shape. A sample object that contains a gas might be a basketball, a soccer ball, or the classroom.
- 11. the gas state
- Gas particles move freely, spread out, and they have little attraction for other gas particles.

After You Read

 Possible answer: A liquid takes the shape but not the volume of its container; a gas takes the shape and volume of its container.

Chapter 2 States Of Matter Answer Key

ML Yell

Chapter 2 States Of Matter Answer Key:

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