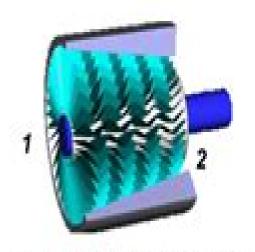
# Conservation of Energy

(Specific form)



$$w = Work = p_2v_2 - p_1v_1 + w_{sh}$$

$$h = Enthalpy = e + pv$$

g = Heat Flow

$$k = Kinectic Energy = \frac{u^2}{2}$$

$$p = pressure$$

e = Internal Energy

u= velocity

v = volume

 $w_{sh} = shaft work$ 

# First Law of Thermodynamics:

$$e_2 - e_1 + k_2 - k_1 = q - w_{sh} - p_2 v_2 + p_1 v_1$$

$$e_2 + p_2 v_2 - e_1 - p_1 v_1 (\frac{u^2}{2})_2 - (u^2/2)_1 = q - w_{sh}$$

$$h_2 + (u^2/2)_2 - h_1 - (u^2/2)_1 = q - w_{sh}$$

define:  $h + (u^2/2) = h_t = \text{Total Enthalpy}$ 

$$h_{t_2} - h_{t_1} = q - w_{sh}$$

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Xiaolong Qi

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