

Concept Review

Section 8-1: Covalent Bonds

Answer the following items in the space provided.

1. What does a molecular bond-forming reaction look like?
2. Why is the H_2 molecule more stable than two separate hydrogen atoms?
3. Explain why the stability difference in H_2 does not show that bond-forming is more energetically favored.
4. What does a molecular bond-forming reaction look like?
5. What is enthalpy change?
6. Describe potential energy changes that occur when two hydrogen atoms approach each other, what forms a covalent bond?
7. In terms of energy, why is the H_2 bond stable?
8. Why is bond length an average value? How is bond energy?
9. Describe the physical properties of molecules that the covalent bond, ionic, and covalent bonds.

Complete each statement below by choosing a term from the following list. Name may be used more than once.

- | | | | |
|---------------------|---------------------|----------|----------------|
| strategy | electron-attraction | electron | ionic covalent |
| electron-attraction | electron | electron | electron |
1. Bond _____ is the force that holds atoms together. Bonding is the force that holds molecules together.
 2. Bond _____ is the force that holds atoms together. Bonding is the force that holds molecules together.
 3. The attraction of an atom for electrons is called the _____.
 4. In _____, electrons are attracted to the nucleus of the atom. In ionic bonding, electrons are attracted to the nucleus of the atom.
 5. In _____, electrons are attracted to the nucleus of the atom. In covalent bonding, electrons are attracted to the nucleus of the atom.

Concept Review Chapter 6 Section Covalent Bonds

Answer Key

William H. Brown, Thomas Poon



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Holt Chemistry Ralph Thomas Myers, 2004 Organic Chemistry Michael B. Smith, 2010-10-18 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules **Principles of Cell Biology** George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the underlying principles that illustrate both how cells function as well as how we study them It identifies 10 specific principles of cell biology and devotes a separate chapter to illustrate each The result is a shift away from the traditional focus on technical details and towards a more integrative view of cellular activity that is flexible and can be tailored to suit students with a broad range of backgrounds CHEMISTRY SILBERBERG, 2003 **Human Biology** Daniel Chiras, 2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course **Class 8-12 Chemistry Questions and Answers PDF** Arshad Iqbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class questions Class 8

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Graham, Lee Warren Wilcox, 2003 This book focuses readers on the function of plants and the role they play in our world The authors emphasize the scientific method to help readers develop the critical thinking skills they need to make sound decisions throughout life This focus on how plants work and the development of critical thinking skills together support the ultimate goal of developing scientific literacy This book is organized around the themes of DNA science global ecology and evolution The key concepts discussed in the book are molecules cells and microbes plant structure and reproduction and plant diversity and the environment For anyone interested in botany plant biology

Biology Cecie Starr, 1994 This streamlined book distills biology's key concepts and connects them to the lives of students with numerous timely applications including compelling new vignettes at the beginning of each chapter Once again Starr created new remarkably clear illustrations to help explain complex biological concepts As with every new edition she continues to simplify and enliven the writing without sacrificing accuracy The author has done a major revision of each chapter so that there is extensive updating and organizational changes to enhance the text's flow As the following features indicate the major thrust of the new edition is to enhance accessibility and further stimulate student interest

General Chemistry Ralph H. Petrucci, William S. Harwood, 1993 *General Chemistry Principles and Modern Applications* is recognized for its superior problems lucid writing and precision of argument This updated and expanded edition retains the popular and innovative features of previous editions including Feature Problems follow up Integrative and Practice Exercises to accompany every chapter Example and Focus On application boxes as well as new Keep in Mind marginal notes Topics covered include atoms and the atomic theory chemical compounds and reactions gases Thermochemistry electrons in atoms chemical bonding liquids solids and intermolecular forces chemical kinetics principles of chemical equilibrium acids and bases electrochemistry representative and transitional elements and nuclear and organic chemistry For individuals interested in a broad overview of chemical principles and applications

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