# CHM 160

CHAPTER 14: WORKSHEET Answers

(Part 1) average rates, rate laws, method of initial rates, integrated rate Laws

Answer the following questions:

Consider the reaction

 $H_2O_2(aq) \rightarrow H_2O(6) + \frac{1}{2} O_2(q)$ 

The initial concentration of hydrogen peroxide is 0.2546 M, and the initial rate of the reaction is  $9.32 \times 10^{-4}$  M/s. What will be  $[H_2O_2]$  at t = 0.58 min?

Answer: 0.222 M

 Consider the hypothetical reaction A + 2B → 3C + 2D At one point in the reaction. [B] = 0.4658 M, and 2.08 minutes later, [B] = 0.3906 M. During this time period. determine a) the average rate of decomposition of B in M-s<sup>-1</sup> b) the average rate of formation of C in M-s1

Answer: a) 6.02×10<sup>-4</sup> M/s; b) 9.03×10<sup>-4</sup> M/s

Consider the reaction:

 $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ 

If the rate of disappearance of H<sub>2</sub> (Δ[H<sub>2</sub>] )is 0.030 mol L<sup>-1</sup> • s<sup>-1</sup>, then calculate the rate of appearance of NH<sub>2</sub> (ANH<sub>2</sub>) during the same time period.

Answer : 0.02 mol L<sup>-1</sup> • s<sup>-1</sup>

2 NOBr(q) → 2NO(g) + Br₂(g)

In the given reaction, which is taking place in a 1.0 L flask, 8.0x10<sup>-4</sup> mol of Br<sub>2</sub> is formed during the first 5 sec of the reaction. What is the rate of disappearance of NOBr during the same period .

Answer: 3.2 x 10<sup>-4</sup> mol/(L.s)

The following kinetic data were obtained for the reaction:  $2 \text{ NO}(g) + O_2(g) \rightarrow 2 \text{ NO}_2(g)$ 

Experiment	[NO]o(M)	[O <sub>2</sub> ]o	Initial rate (M/s)
1	0.010	0.010	2.5 x 10 <sup>-5</sup>
2	0.020	0.010	1.0 x 10 <sup>-4</sup>
3	0.010	0.020	5.0 x 10 <sup>-5</sup>
4	0.015	0.0050	?

- a) Write the rate law for the reaction.
- b) Calculate the rate constant k.
- c) Use the data to predict the reaction rate for experiment 4.

Answer: a. rate = k[NO]2[O2] b. k = 2.5 x 101 M2s1 c. 2.8x 105 Ms1

# **Core Teaching Resources Chemistry Answers Chapter 14**

**Robert R. Redfield** 

# **Core Teaching Resources Chemistry Answers Chapter 14:**

Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents **Resources in Education** ,1995 **Organic Chemistry** Francis A. Carey, 2000 A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless

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