

Name: _____

Diffusion and Osmosis

Lab Data and Observations (continued)

Part B. Diffusion through a membrane

- At the end of the experiment (Part B):

What is the color of the solution (white/orange/black) in the bag? _____

Molecules/ions of which three compounds are now present in the bag? _____

- Based on your observations tell molecules of which compound moved
In which direction did they move (from the beaker into the bag or from the bag into the beaker)? _____

Why did they move in that direction? _____

- Molecules of which substance did **not** move through a membrane (despite of the concentration difference) _____ (one word)

Why did not they move? _____

If you are not sure how to answer any of these questions re-read the foreword to the Part B procedure.

Part C. Osmosis

Mass of a bag (in grams) - measured					
Bag	0 min (beginning of the experiment)	10 min	20 min	30 min	40 min
	2.58 <i>Do not use data from this row - they are given only as an example how to record the data</i>	2.98	3.33	3.75	4.14
1	9.56	9.70	9.45	9.60	9.72
2	10.20	11.35	12.40	13.05	14.23
3	13.46	15.91	17.06	17.06	19.26
4	11.31	10.28	9.28	9.11	6.02

Changes in mass (in grams) - calculated and to be plotted on the next page

How to calculate it: **always subtract the mass at time 0 in the table above from the mass at a given time**

Bag	0 min	10 min	20 min	30 min	40 min
0*	0 <i>Do not use data from this row - they are given only as an example how to calculate the data in this table</i>	0.40 (2.98-2.58)	0.75 (3.33-2.58)	1.17 (3.75-2.58)	1.56 (4.14-2.58)
1	0	0.14	-0.11	0.04	0.16
2	0	1.15	2.2	2.85	4.03
3	0	2.45	3.6	5.1	6.8
4	0	-1.03	-2.2	-3.25	-5.29

Diffusion Through A Membrane Lab Answers

Gregory Scott Hunter



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