

**Dimensional Analysis Problems
Chemistry I**

1. Convert 25 miles per gallon to kilometers per liter. **(11km/L)**

2. What is the volume in Liters of a room that is 10 meters wide by 15 meters long and 8 ft high? **(4X10⁵ L)**

3. If the room in problem 2 is filled with air that has a density of 1.19 g/L , what is the mass of air in the room, in Kilograms? **(500kg)**

4. A leaky faucet drips at a rate of 3.5 mL per second. How many gallons of water per day are wasted? **(80 gal)**

5. An empty vial weighs 31.45 grams. If the vial weighs 179.56 g when filled with liquid mercury, what is its volume? The specific gravity of Hg liquid is 13.53. **(10.95mL)**

6. A handbook gives the density of calcium as 1.54g/mL. A student measures the density of calcium to be 1.25g/mL in the lab. What is the % error based on the lab measurement? **(19%)**

Dimensional Analysis Practice Problems Chemistry

Lingjun Ying



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Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H. Langley, 2014-03-11 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time *The Journal of Physical Chemistry* ,1924 **Chemical Problem-solving by Dimensional Analysis** Arnold B. Loebel, 1974 *Ebook: Introductory Chemistry: An Atoms First Approach* Burdge, 2016-04-16 Ebook Introductory Chemistry An Atoms First Approach **Ebook: Chemistry** Julia Burdge, 2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text *Addison-Wesley Chemistry* ,1990 Chemical Problem Solving Using Dimensional Analysis Robert Nakon, 1978 *Chemistry* Jason Overby, Raymond Chang, 2024 The fifteenth edition continues a long tradition of providing a firm foundation in the concepts of chemical principles while instilling an appreciation of the important role chemistry plays in our daily lives We believe that it is our responsibility to assist both instructors and students in their pursuit of this goal by presenting a broad range of chemical topics in a logical format At all times we strive to balance theory and application and to illustrate principles with applicable examples whenever possible Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change Prentice Hall Chemistry ,2000 **Student Study Guide to Accompany Chemistry** Kenneth J. Hughes, Paul B. Kelter, 2002-05 Reviews essential points from the text presents problems and hints to assist the student in solving them Solutions to selected problems in the end of chapter problem sets are included By Kenneth Hughes of the University of Wisconsin Madison **Preparatory Chemistry** Howard Stephen

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