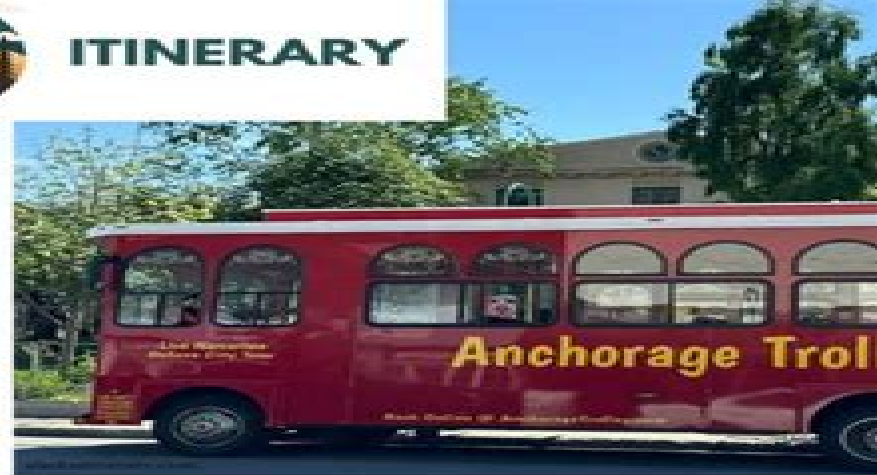


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Wilderness Guide is the book for anyone interested in adventure travel in Alaska and is perfect for the backpacker business traveler or armchair traveler Living up to its reputation as the most comprehensive guide available The Alaska Wilderness Guide provides detailed descriptions of 250 bush communities 85 navigable rivers and 100 parks refuges monuments wilderness areas and special features in Alaska Travel Books Worldwide says The Alaska Wilderness Guide is a fine gazetteer filled with descriptions of places that few if any other guidebooks include The 7th edition includes a directory of adventure travel businesses fishing hunting guide services boat charters wilderness lodges and air taxis in the Bush **Gordon's Guide to Adventure Vacations** Timothy E. Gordon, Ana Bela Oliveira, 1998 This book is for those in search of fun Cowboy up at an authentic dude ranch Horse pack with your spouse in Montana s rugged Bitterroot Wilderness Bicycle through the beautiful vineyards of Napa Valley Italy or France Drive a covered wagon pulled by a team of horses through the Teton wilderness of Wyoming Plunge down the majestic Grand Canyon s Lava Falls the world s biggest whitewater Feel the exhilaration kayaking among Orcas on Alaska s Inside Passage or paddling the myriad of islands in the crystal clear waters of the Belize Features 700 recreation providers DiRoNA 2005 Guide to Distinguished Restaurants of North America DiRoNA, 2004 The 2005 guide tells you all you need to know from dress codes to parking options at each restaurant along with photos of ambiance and specialty dishes Buyers of the book who have less than an exceptional fine dining experience at a DiRoNA restaurant may obtain up to a 200 refund on their restaurant bill **Compass American Guides: Alaska, 4th Edition** , 2005-05 An illustrated guide to Alaska **Woodall's Campground Directory** , 1996

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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