

CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple Choice Questions

1. Which of the statements about the reaction below are incorrect ?

$$2\text{PbO}(s) + \text{C}(s) \longrightarrow 2\text{Pb}(s) + \text{CO}_2(g)$$

- (i) Lead is getting reduced
- (ii) Carbon dioxide is getting oxidised
- (iii) Carbon is getting oxidised
- (iv) Lead oxide is getting reduced
- (a) (i) and (ii)
- (b) (i) and (iii)
- (c) (i), (ii) and (iii)
- (d) all of these.

2. $\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$

The above reaction is an example of a—

(2024)

- (a) combination reaction
- (b) double displacement reaction
- (c) decomposition reaction
- (d) displacement reaction.

3. What happens when dilute hydrochloric acid is added to iron fillings ? (2022)

- (a) Hydrogen gas and iron chloride are produced
- (b) Chlorine gas and iron hydroxide are produced
- (c) No reaction takes place
- (d) Iron salt and water are produced.

In the reaction $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$ Chlorine is a—

- (a) Reducing agent
- (b) Oxidizing agent
- (c) Catalyst
- (d) Inert medium.

Rancidity can be prevented in food items by—

- (a) Mixing antioxidant
- (b) Keeping in refrigerator
- (c) Keeping away from light
- (d) All of these.

Which among the following is not an oxidizing agent ?

- (a) Oxygen
- (b) Conc. Sulphuric acid
- (c) Chlorine
- (d) Hydrogen.

Which among the following gases are used to keep the oily food items fresh ?

- (a) Carbon dioxide and Oxygen
- (b) Nitrogen and helium
- (c) Helium and Oxygen
- (d) Nitrogen and Oxygen.

Chemically rust is—

- (a) Hydrated ferrous oxide
- (b) Ferric oxide
- (c) Hydrated ferric oxide
- (d) None of these.

An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

- (a) Respiration
- (b) Digestion
- (c) Excretion
- (d) None of these.

Chapter 1 chemical Periodicity Answers

Rob Ritchie



Chapter 1 chemical Periodicity Answers:

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Table of Contents Chapter 1chemical Periodicity Answers

1. Understanding the eBook Chapter 1chemical Periodicity Answers
 - The Rise of Digital Reading Chapter 1chemical Periodicity Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 1chemical Periodicity Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 1chemical Periodicity Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 1chemical Periodicity Answers
 - Personalized Recommendations
 - Chapter 1chemical Periodicity Answers User Reviews and Ratings
 - Chapter 1chemical Periodicity Answers and Bestseller Lists
5. Accessing Chapter 1chemical Periodicity Answers Free and Paid eBooks
 - Chapter 1chemical Periodicity Answers Public Domain eBooks
 - Chapter 1chemical Periodicity Answers eBook Subscription Services
 - Chapter 1chemical Periodicity Answers Budget-Friendly Options

6. Navigating Chapter 1chemical Periodicity Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 1chemical Periodicity Answers Compatibility with Devices
 - Chapter 1chemical Periodicity Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 1chemical Periodicity Answers
 - Highlighting and Note-Taking Chapter 1chemical Periodicity Answers
 - Interactive Elements Chapter 1chemical Periodicity Answers
8. Staying Engaged with Chapter 1chemical Periodicity Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 1chemical Periodicity Answers
9. Balancing eBooks and Physical Books Chapter 1chemical Periodicity Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 1chemical Periodicity Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 1chemical Periodicity Answers
 - Setting Reading Goals Chapter 1chemical Periodicity Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 1chemical Periodicity Answers
 - Fact-Checking eBook Content of Chapter 1chemical Periodicity Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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