

# > Workbook answers

## Chapter 1

### Exercise 1.1

1. A solid line is fixed **monomer volume** (slope) & liquid line is fixed **mass** but its **slope** changes for that of the container (depends if it is placed). It goes from one fixed **slope** or **volume** to give completely little information about it to be so.

2.



(a)



(b)

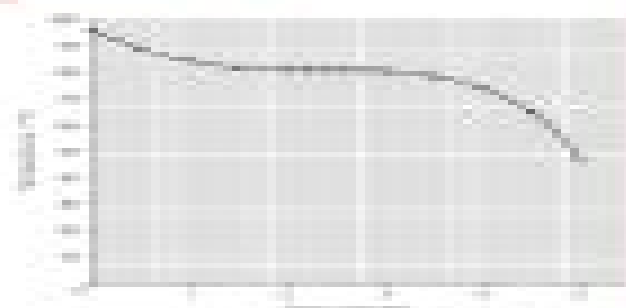


(c)

3.
  - a. B
  - b. C
  - c. A
  - d. D
4. In freezing (solidification):
  - a. cooling
  - b. crystallization
  - c. supercooling or freezing
5.
  - a. fusion
  - b. solid-solid transition
  - c. melting
  - d. solid
  - e. Two examples of polymers used by humans: The polymer of temperature sensitive hydrogel of crystallization.

### Exercise 1.2

1.



2. The constant **supercooled** region can not be the plot of the water freezing as that the higher temperature would be reached.
3. The influence is freezing **enthalpy** & **entropy** from liquid to solid.
4. The temperature **supercooled** freezing energy is high, **enthalpy** & **entropy** **solidification** the molecules are going out from (they) stop moving from place to place until become organized in a structured lattice arrangement. In the solid the molecules can not move without other fixed points. The heat released by the formation of new **hydrogen bonds** between the particles keeps the temperature constant until the substance is solid.

# Chapter 1chemistry Workbook Answers

**Burns**



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