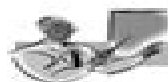


CHEMICAL REACTIONS AND EQUATIONS



INTRODUCTION

A substance can undergo a change by two ways -

- (i) **Physical changes** - This is a change in physical properties such as shape, size, colour and state of a substance.

- No new substance is formed.
- A physical change is generally reversible.

e.g. Boiling of water, condensation of water vapour, melting of ice, dissolution of sugar in water etc.

- (ii) **Chemical changes** - This is a change in which one or more new substances with new properties are formed.

- Original substance loses their nature and identity.
- Rearrangement of atoms take place.
- A chemical change can be reversible or irreversible.

A chemical change is also known as chemical reaction. e.g. Rusting of iron, souring of milk, burning of fuel, cooking of food, respiration, digestion of food in our body, ripening of fruits etc.

Thus, the process involving a chemical change is called a chemical reaction. The presentation of a chemical change in the equation is known as chemical equation.

Activity 1.

Take a magnesium ribbon. Clean it by rubbing with sand paper.

- (a) **Why should magnesium ribbon be cleaned before burning in air ?**

Hold it with a pair of tongs. Burn it in the flame of a burner. Collect the ash so formed in a watch glass.

Precaution - Burn magnesium ribbon by keeping it as far as possible from your eyes.

- (b) **What is your inference?**

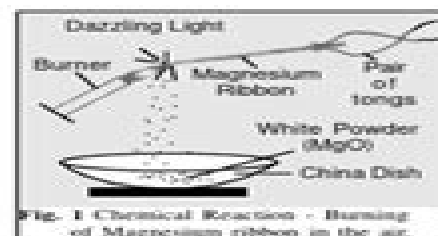
Objective-To understand a chemical reaction.

Inference -

- (a) Magnesium ribbon (Mg) is cleaned before burning to remove the layer of magnesium oxide, dirt etc. from its surface. It forms when magnesium ribbon remains exposed to moist air. Such layer obstructs burning of magnesium.

- (b) On burning in air, magnesium ribbon gives dazzling white flame and changes into a white powder (magnesium oxide, MgO) as a result of reaction between magnesium and oxygen present in air.

Conclusion - This is a chemical reaction as original substance (Mg ribbon) loses its nature and identity.



CHARACTERISTICS OF CHEMICAL REACTION

When a chemical reaction occurs, following observations help us to determine whether a chemical reaction has taken place.

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Patrick Vollmar



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