

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: $\uparrow \text{temp} = \uparrow \text{speed}$

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

↓
b/c molecules moving faster

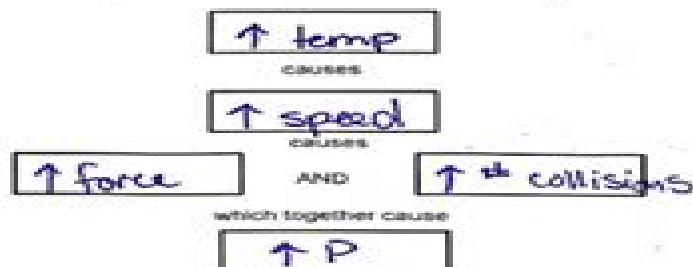
$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

Chapter 1the Behavior Of Gases Packet Answers

N Colangelo



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Chemistry John Olmsted, Gregory M. Williams, 1997 Textbook outlining concepts of molecular science **Holt Chemistry**
Ralph Thomas Myers, 2004 *8th Congress on Electronic Structure: Principles and Applications (ESPA 2012)* Juan J. Novoa, Manuel F. Ruiz López, 2013-10-16 This volume collects research findings presented at the 8th Edition of the Electronic Structure Principles and Applications ESPA 2012 International Conference held in Barcelona Spain on June 26 29 2012 The contributions cover research work on methods and fundamentals of theoretical chemistry chemical reactivity bimolecular modeling and materials science Originally published in the journal *Theoretical Chemistry Accounts* these outstanding papers are now available in a hardcover print format as well as a special electronic edition This volume provides valuable content for all researchers in theoretical chemistry and will especially benefit those research groups and libraries with limited access to the journal **Combustion and Incineration Processes** Walter R. Niessen, 2010-06-22 In our throwaway society with landfills filled to capacity interest in incineration and conversion based waste management technologies continues to grow Increasing net waste generation rates within U S metropolitan centers skyrocketing transportation costs for waste hauling and the enticement of increased electrical revenues from green p **Key-words-in-context Title Index**, 1962 Energy Research Abstracts, 1986 *Scientific and Technical Aerospace Reports*, 1991 El-Hi Textbooks and Serials in Print, 1996 R R Bowker Publishing, Bowker, 1996-02 *The Software Encyclopedia*, 1986 **Government Reports Annual Index**, 1989 **Government Reports Index**, 1974 *Early Years*, 1984 Government Reports Announcements & Index, 1994 **Resources in Education**, 1979 The Elementary School Library Collection, Phases 1-2-3, 1976 **The Elementary School Library Collection**, 1976 Library of Congress Catalog Library of Congress, 1971 **Addison-Wesley Science Insights**, 1996 **Science Insights**, 1999 **Physics Briefs**, 1992

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