

Chapter 2 Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. All of the following are steps in the scientific method *except*
- observing and recording data.
 - forming a hypothesis.
 - discarding data inconsistent with the hypothesis.
 - developing a model based on experimental results.
- _____ 2. Which of the following observations is quantitative?
- The liquid turns blue litmus paper red.
 - The liquid boils at 100°C .
 - The liquid tastes bitter.
 - The liquid is cloudy.
- _____ 3. Which of the following observations is qualitative?
- A chemical reaction was complete in 2.3 seconds.
 - The solid had a mass of 23.4 grams.
 - The pH of a liquid was 5.
 - Salt crystals formed as the liquid evaporated.
- _____ 4. Which of these is *not* an SI base unit?
- kilogram
 - second
 - liter
 - Kelvin
- _____ 5. The symbol mm represents
- micrometer.
 - millimeter.
 - milliliter.
 - meter.
- _____ 6. Which of these metric units is used to measure mass?
- m
 - mm
 - g
 - L
- _____ 7. The SI base unit for length is the
- meter.
 - millimeter.
 - centimeter.
 - kilometer.
- _____ 8. The relationship between the mass m of a material, its volume V , and its density D is
- $D = mV$.
 - $D = V/m$.
 - $D = m/V$.
 - $D = m + v$.
- _____ 9. The density of aluminum is 2.70 g/cm^3 . What is the mass of a solid piece of aluminum with a volume of 1.50 cm^3 ?
- 0.556 g
 - 1.80 g
 - 4.05 g
 - 4.20 g
- _____ 10. The density of pure diamond is 3.5 g/cm^3 . What is the volume of a diamond with a mass of 0.25 g?
- 0.071 cm^3
 - 0.875 cm^3
 - 3.75 cm^3
 - 14 cm^3
- _____ 11. 100 milliliters is equivalent to

Chapter 2 Chemistry Test

Thomas Evangelist



Chapter 2 Chemistry Test:

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