

# Chapter 20: Acids and Bases

## -- Describing Acids and Bases --

### Names and Formulas of Acids

- When the anion name ends in “-ate”
  - the acid is named by replacing “-ate” with “-ic” and adding the word “*acid*” to the end.
- Examples
  - $\text{H}_2\text{SO}_4$ : sulfuric acid (since it contains the sulfate ion)
  - $\text{HNO}_3$ : nitric acid (since it contains the nitrate ion)

## Chapter 20 Describing Acids And Bases

**Patricia A. Potter, Anne G.  
Perry, Patricia A. Stockert, Amy Hall**

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