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Guided Practice - Acids & Bases

Properties of Acids and Bases

- 1. Match the following terms with their descriptions:
 - is. Airhenius Acid ______ the substance formed when a proton is added to
 - a base
 - Arrhenius Base ______ the remaining substance when a proton is lost
 - from an acid
 - Conjugate Acid ______ a substance that produces OH in solution
 - d. Conjugate Base a substance producing H* in solution
 - e. Bonsted-Lowry Acid ______ a proton acceptor
 - f. Bronsted-Lowry Base a proton donor
- Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, liabeling each species correctly.
 - a HSON SON
 - b. Hibr. Bro-
 - e. Hypout Pour
 - d. H890's 190"
- 3. Complete the following table regarding acid strength.

Property	Strong Acid	Wesk Acid
Substation		
Describe Dissociation		
Species present in solution		
Relative strength of corpugate base compared to mater		
Equation showing acid's reaction to mater		
Paragraphy List in acids		

Chapter 1review Acids Bases Answer Key

Patrick Vollmar

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