

HOLT MODERN CHEMISTRY – CHAPTER 1 REVIEW

*The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test. ***** Test items will be recall, examples, and/or application of this content. ******

OUTCOMES

- 1.1: Distinguish between types of research – basic, applied, and technological development. (F & PK)
- 1.2: Distinguish between the physical and chemical properties of matter. (F & PK)
- 1.2: Distinguish among homogeneous and heterogeneous mixtures, elements, and compounds. (F & PK)
- 1.3: List properties of metals and non-metals. (T & R)
- 2.3: Determine significant figures. (F & PK)

1.1: CHEMISTRY IS A PHYSICAL SCIENCE

- **Vocabulary**
 - **chemistry** – the scientific study of the composition, structure, and properties of matter and the changes that matter undergoes
 - **chemical** – any substance with a defined composition
- **Chapter Highlights**
 - Chemistry is the study of the composition, structure, and properties of matter and the changes that matter undergoes.
 - A chemical is any substance that has a definite composition or is used or produced in a chemical process.
 - Basic research is carried out for the sake of increasing knowledge. Applied research is carried out to solve practical problems. Technological development involves the use of existing knowledge to make life easier or more convenient.

1.2: MATTER AND ITS PROPERTIES

- **Vocabulary**
 - **matter** – anything that has mass and takes up space
 - **mass** -- a measure of the amount of matter in an object
 - **element** – a substance that cannot be separated or broken down into simpler substances by chemical means; all atoms of an element have the same atomic number
 - **atom** – the smallest unit of an element that maintains the chemical properties of that element
 - **compound** -- a substance made up of atoms of two or more different elements joined by chemical bonds
 - **extensive property** -- a property that depends on the extent or size of a system
 - **intensive property** -- a property that does not depend on the amount of matter present, such as pressure, temperature, or density
 - **physical property** -- a characteristic of a substance that does not involve a chemical change, such as density, color, or hardness
 - **physical change** -- a change of matter from one form to another without a change in chemical properties
 - **change of state** -- the change of a substance from one physical state to another
 - **solid** -- the state of matter in which the volume and shape of a substance are fixed
 - **liquid** -- the state of matter that has a definite volume but not a definite shape
 - **gas** -- a form of matter that does not have a definite volume or shape
 - **plasma** -- in physical science, a state of matter that starts as a gas and then becomes ionized; it consists of freemoving ions and electrons and it takes on an electric charge
 - **chemical property** -- a property of matter that describes a substance's ability to participate in chemical reactions
 - **chemical change** -- a change that occurs when one or more substances change into entirely new substances with different properties
 - **chemical reaction** -- the process by which one or more substances change to produce one or more different substances
 - **reactant** -- a substance or molecule that participates in a chemical reaction
 - **product** -- a substance that forms in a chemical reaction

Chapter 1review Modern Chemistry

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Chapter 1review Modern Chemistry:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Modern Chemistry* Nicholas D. Tzimopoulos, 1993 *Introduction to Modern Inorganic Chemistry* Joseph William Mellor, 1914 **Modern inorganic chemistry** Joseph William Mellor, 1918 **Solutions Guide for Introductory Chemistry** James F. Hall, 1996 Modern Physical Organic Chemistry Eric V. Anslyn, Dennis A. Dougherty, 2006 Making explicit the connections between physical organic chemistry and critical fields such as organometallic chemistry materials chemistry bioorganic chemistry and biochemistry this book escorts the reader into an area that has been thoroughly updated in recent times **Leaves of Healing**, 1928 Modern Colorants: Synthesis and Structure A.T. Peters, H.S. Freeman, 2014-02-19 Although the research activities of dyestuff chemists worldwide have been influenced to a great extent in recent years by the need to respond to a variety of environmental issues associated with the manufacture and application of synthetic dyes and pigments a significant level of targeted research continues to be devoted to new chemistry aimed at enhancing the technical properties of dyes in commerce This book is a presentation of various aspects of basic research conducted during the past decade but not reported in the recent review literature The coverage herein is unique in that it emphasizes systematic approaches commonly utilized in the design and synthesis of dyes and pigments and the required intermediates While it is well known that certain transition metals are important in the synthesis of technically viable metallized dyes for polyamide and protein fibers these metals are demonstrated in Chapter 1 also to be effective agents in the regiospecific placement of substituents into azo compounds The scope and limitations of this chemistry are presented In other synthetic work a description of the different processes employed to produce the major families of reactive dyes is presented In Chapter 4 special attention is given to reactive dyes containing more than one reactive group and to the more recent developments in the field The two chapters which follow provide a review of the recent literature pertaining to novel chromophores and dyes for the D2T2 process respectively

Solutions Guide, Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry, Fourth Edition, Zumdahl James F. Hall, 2000 *The Chemical Trade Journal and Oil, Paint and Colour Review*, 1891 Chemistry and Manufacture of Writing and Printing Inks New York Public Library, 1926 **The Christian Life**, 1883 Water and Gas Review, 1899 Fundamentals of Sustainable Chemical Science Stanley E. Manahan, 2009-03-10 Written by Stanley Manahan Fundamentals of Sustainable Chemical Science has been carefully designed to provide a basic introduction to chemistry including organic chemistry and biochemistry for readers with little or no prior background in the subject Manahan bestselling author of many environmental texts presents the material in a practical *The Yale Alumni Weekly*, 1908 Ess Chem Probs Study Guide David Margolese, David W. Oxtoby, 2004-04 Principles of Modern Chemistry David W. Oxtoby, Norman H. Nachtrieb, 1990 PRINCIPLES OF MODERN CHEMISTRY has dominated the honors and high mainstream general chemistry courses and is considered the standard for the course The fifth edition is a substantial revision

that maintains the rigor of previous editions but reflects the exciting modern developments taking place in chemistry today Authors David W Oxtoby and H P Gillis provide a unique approach to learning chemical principles that emphasizes the total scientific process from observation to application placing general chemistry into a complete perspective for serious minded science and engineering students Chemical principles are illustrated by the use of modern materials comparable to equipment found in the scientific industry Students are therefore exposed to chemistry and its applications beyond the classroom This text is perfect for those instructors who are looking for a more advanced general chemistry textbook

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