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THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the item.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed mass of gas at constant pressure
2. The volume of a fixed mass of gas at constant pressure as the absolute temperature is changed
3. The ratio of the mass of a gas to its volume, independent of the actual mass or volume used
4. Temperature at which a fixed mass of gas exerts zero pressure at fixed volume
5. A measure of how much the volume of a fixed mass of gas changes
6. The a combination of gas pressure, temperature, and volume of the gas mathematically with pressure
7. An ideal gas that obeys the gas laws perfectly
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of how much a fixed mass of gas expands when heated
10. A fixed volume of gas

Column II

- a. pressure, p
- b. ideal gas constant, R
- c. density
- d. compressibility
- e. Boyle's law
- f. Boyle's equation
- g. Charles' law of expansion, pressure
- h. volume
- i. compressibility
- j. coefficient of expansion

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Expanding one volume of a substance gives the same work under varying temperature conditions as expanding a) one volume of another substance b) two volumes of the same substance c) one volume of the same substance d) two volumes of the same substance

Chapter 1the Behavior Of Gases Test B

Mark Steyvers



Chapter 1the Behavior Of Gases Test B:

Introductory Theoretical Chemistry Grover Howard Cartledge,1929 *Fundamentals of Natural Gas Processing, Third Edition* Arthur J. Kidnay,William R. Parrish,Daniel G. McCartney,2019-10-01 Offering indispensable insight from experts in the field *Fundamentals of Natural Gas Processing Third Edition* provides an introduction to the gas industry and the processes required to convert wellhead gas into valuable natural gas and hydrocarbon liquids products including LNG The authors compile information from the literature meeting proceedings short courses and their own work experiences to give an accurate picture of where gas processing technology stands today as well as to highlight relatively new technologies that could become important in the future The third edition of this bestselling text features updates on North American gas processing and changing gas treating requirements due to shale gas production It covers the international nature of natural gas trade LNG economics and more To help nonengineers understand technical issues the first 5 chapters present an overview of the basic engineering concepts applicable throughout the gas oil and chemical industries The following 15 chapters address natural gas processing with a focus on gas plant processes and technologies The book contains 2 appendices The first contains an updated glossary of gas processing terminology The second is available only online and contains useful conversion factors and physical properties data Aimed at students as well as natural gas processing professionals this edition includes both discussion questions and exercises designed to reinforce important concepts making this book suitable as a textbook in upper level or graduate engineering courses *Harcourt Science: Teacher's ed., life science units A and B* ,2005 *Focus on physical science* Charles H. Heimler,1989 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Chemistry* Steven S. Zumdahl,Susan A. Zumdahl,2000 Zumdahl s conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations *Harcourt Science* ,2000 *Foundations of Chemistry* Eleanor Dantzler Siebert,1982 *Prentice Hall Chemistry* ,2000 **A Laboratory Outline of General Chemistry** Alexander Smith,1899 **Laboratory Manual of General Inorganic Chemistry** Mayce Cannon Sneed,Raymond Eller Kirk,1926 *Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and*

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